

**2008 SUMMARY REPORT
of
Countryside Lake**

Lake County, Illinois

Prepared by the

**LAKE COUNTY HEALTH DEPARTMENT
ENVIRONMENTAL HEALTH SERVICES
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LAKE FACTS

Lake Name:	Countryside Lake
Historical Name:	None
Nearest Municipality:	Mundelein
Location:	T44N, R10E, Sections 26,27,34,35
Elevation:	785.0 feet mean sea level
Major Tributaries:	Indian Creek
Watershed:	Des Plaines River
Sub-watershed:	Indian Creek
Receiving Waterbody:	Des Plaines River
Surface Area:	141.8 acres
Shoreline Length:	3.9 miles
Maximum Depth:	9.0 feet
Average Depth:	5.8 feet
Lake Volume:	820.9 acre-feet
Lake Type:	Impoundment
Watershed Area:	1780.0 acres
Major Watershed Land Uses:	Single Family, Forest and Grassland
Bottom Ownership:	Private
Management Entities:	Countryside Lake Association
Current and Historical Uses:	Fishing, no wake boating, swimming
Description of Access:	No public access

Countryside Lake was chosen to be one of seven “sentinel” lakes in the county that the Lakes Management Unit (LMU) will be monitoring annually for five years, beginning with the 2005 season. This report summarizes the water quality sampling results and aquatic plant surveys conducted in 2008 on Countryside Lake. Similar reports have been written on data collected in 2000, 2005, 2006, and 2007 and are available from the LMU at (847) 377-8030 or on the web at (<http://www.lakecountyil.gov/Health/want/LakeReports.htm>). 2009 will be the final year of the annual “sentinel” lake monitoring. A comprehensive summary report detailing all five years will be completed and available for review in 2010.

SUMMARY OF WATER QUALITY

Water samples were collected monthly from May through September at the deepest point in the lake (Figure 1; Appendix A). Countryside Lake was sampled at depths of three feet and six feet (May and June only), and the samples were analyzed for various water quality parameters (Appendix C). In addition, the LMU’s beach sampling program has included the Countryside Lake Association (CLA) beach since 1988. In 2008 there was one recommended swim ban.

Countryside Lake was well mixed throughout the season due to the lake being shallow and easily mixed by wind. This mixing was reflected in the dissolved oxygen (DO) concentrations, as well as the other water quality parameters. The concentrations from the epilimnion were similar to the hypolimnion, therefore only the data from the epilimnion will be discussed. DO concentrations of at least 5.0 mg/L were recorded in Countryside Lake in 2008 from the water’s surface down to near eight feet during all months (Appendix B). In June, July, and August the DO concentration was below 5.0 mg/L at depths greater than eight feet likely due to the decomposition of plants and algae from the monthly chemical treatments. Using data from a bathymetric map created by STS consultants in 2001, the volume of the lake greater than eight feet accounts for 1.3% of the total lake volume. Thus, the volume of the lake with a DO concentration below the amount necessary to support aquatic life was minimal.

Secchi disk depth (water clarity) averaged 3.36 feet during 2008, which was above the Lake County median of 3.12 feet (Appendix D). This was a decrease from the previous averages of 5.38 feet in 2007, 5.07 feet in 2006, and 4.09 feet in 2005 (Table 1). Countryside Lake has participated in the Illinois Environmental Protection Agency’s (IEPA) Volunteer Lake Monitoring Program (VLMP) since 2001. The VLMP Secchi depth averaged 4.46 feet over the past seven years ranging from 2.97 feet (2002) to 8.08 feet (2007, Figure 2). During 2007 VLMP data was only collected during May and June which are the two months that tend to have better water clarity. If 2007 was omitted from the average, the VLMP Secchi depth average was 3.84 feet. Water clarity is related to the amount of total suspended solids (TSS) concentration in the water column. As the TSS increased, the water clarity decreased (Figure 3). The 2008 average TSS concentration of 6.2 mg/L was an increase from the 2007 (4.7 mg/L) and 2005 (4.0 mg/L) average, but a decrease from the 2006 (6.7 mg/L) average. The 2008 average TSS was below the Lake County median of 8.2 mg/L. The variability of the water clarity and TSS could be a result of fluctuations in algae blooms from year to year or a result of the amount of precipitation during a given year. According to the CLA rain gauge nearly 23 inches of rain fell from May through September 2008. Prior to the September sampling, nearly nine inches of rain

was recorded at the CLA rain gauge. This rain event caused a decrease in the Secchi depth (1.15 feet) and an increase in the TSS (11.0 mg/L). As part of the monthly sampling, water level was measured at the boat launch near the outlet. The lake level increased by 1.3 inches from May through June and then decreased 4.5 inches from June to July. The level then dropped another 2.0 inches from July to August. No reading was recorded during the September sampling event.

Data collected by the IEPA, LMU, and CLA has shown negative impacts on the Countryside Lake watershed due to runoff from an upstream development. CLA has also been monitoring the lake's turbidity, TP, TSS, and total organic carbon after significant rain events at six locations around the lake. They continue to see high levels entering the lake from Indian Creek, however not as high as when the land upstream was under development.

Total phosphorous (TP) increased from 2007 but was still below the Lake County median of 0.065 mg/L. The average TP concentrations for 2005, 2006, 2007, and 2008 were 0.051 mg/L, 0.079 mg/L, 0.033 and 0.062 mg/L, respectively. This increase in TP was likely the result of a decrease in plant density from the previous year as aquatic plants utilize phosphorous. In lakes that do not stratify, like Countryside Lake, phosphorus can be released from sediment through biological/mechanical processes or from plants/algae as they die. The trophic state of Countryside Lake, in terms of its phosphorus concentration, from 2005 through 2008 was eutrophic, with TSIp scores of 60.9, 67.1, 54.7, and 63.6, respectively. Countryside Lake was 75th out of 163 lakes in Lake County (Table 2).

The IEPA has assessment indices to classify Illinois lakes for their ability to support aquatic life and recreational uses. The guidelines consider several aspects, such as water clarity, phosphorus concentrations (for the trophic state index), and aquatic plant coverage. According to this index, Countryside Lake provided *Full* support of aquatic life and *Partial* support of recreational activities due to the high TP concentration, and provided *Partial* overall use in 2008.

The Lake County median conductivity reading for near surface samples was 0.8195 milliSiemens/cm (mS/cm). During 2008, the average conductivity reading for Countryside Lake was lower at 0.6944 mS/cm. This was down from the 2005 and 2006 averages of 0.7546 mS/cm and 0.7196 mS/cm, but up from the 2007 average of 0.6842 mS/cm. The 2005 and 2006 averages were likely higher due to lower total precipitation, which minimized any "flushing" and through evaporation the lake volume decreased concentrating the dissolved ions. In 2007, the increased water level increased the volume of the lake and allowed water to flow over the dam. The 2008 average chloride (Cl⁻) concentration in Countryside Lake was lower than the Lake County median of 166 mg/L, with a seasonal average of 121 mg/L. This was down from the 2007 average of 128 mg/L and the 2006 average of 133 mg/L and up from the 2005 average of 122 mg/L. Road salt used for winter de-icing was the probable source of Cl⁻. The Illinois Environmental Protection Agency (IEPA) standard for chloride is 500 mg/L. Once values exceed this standard, the waterbody is deemed to be impaired, thus impacting aquatic life. A study done in Canada reported 10% of aquatic species were harmed by prolonged exposure to Cl⁻ concentrations greater than 220 mg/L. Additionally, shifts in algal populations were associated with Cl⁻ concentrations as low as 12 mg/l. Therefore, it is important to keep the use of road salts to a minimum within the watershed, particularly in areas with new development.

Figure 1. Water quality sampling site on Countryside Lake.

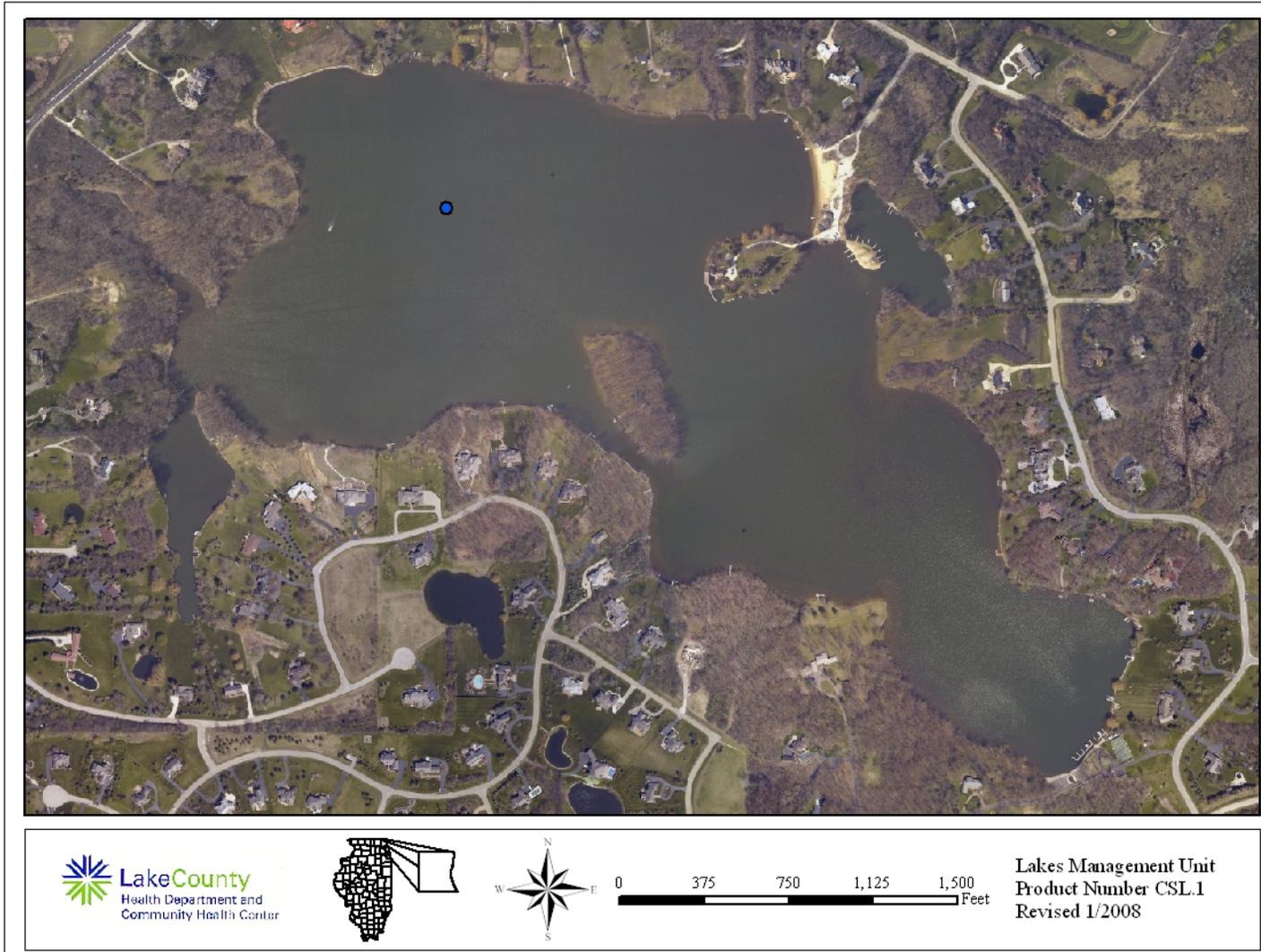


Table 1. Water quality data for Countryside Lake, 2005 – 2008

2008	Epilimnion														
DATE	DEPTH	ALK	TKN	NH ₃ -N	NO ₂ +NO ₃ -N	TP	SRP	Cl ⁻	TSS	TS	TVS	SECCHI	COND	pH	DO
14-May	3	129	0.61	<0.1	<0.05	0.029	<0.005	125	2.1	425	98	6.46	0.7177	8.87	10.13
11-Jun	3	133	0.64	<0.1	<0.05	0.038	0.007	117	1.6	400	90	2.46	0.6980	8.58	7.61
09-Jul	3	132	1.14	<0.1	<0.05	0.067	<0.005	120	8.4	415	115	3.94	0.6892	9.18	9.94
13-Aug	3	137	1.15	<0.1	<0.05	0.093	<0.005	124	8.1	415	103	2.79	0.6857	8.86	8.68
10-Sep	3	143	1.30	<0.1	<0.05	0.081	<0.005	119	11.0	396	99	1.15	0.6816	8.52	9.13
Average		135	0.97	<0.1	<0.05	0.062	0.007 ^k	121	6.2	410	101	3.36	0.6944	8.80	9.10
2007	Epilimnion														
DATE	DEPTH	ALK	TKN	NH ₃ -N	NO ₂ +NO ₃ -N	TP	SRP	Cl ⁻	TSS	TS	TVS	SECCHI	COND	pH	DO
08-May	3	167	0.54	<0.1	<0.05	0.025	<0.005	134	3.3	502	98	5.25	0.7510	8.19	8.57
12-Jun	3	158	0.66	<0.1	<0.05	0.014	<0.005	139	<1.0	512	137	9.20 ^a	0.7110	8.40	9.04
10-Jul	3	152	0.74	<0.1	<0.05	0.024	<0.005	142	3.2	508	137	5.32	0.7270	8.38	6.69
07-Aug	3	131	0.71	<0.1	<0.05	0.030	<0.005	137	3.2	457	111	4.26	0.6910	8.44	7.45
11-Sep	3	151	1.00	<0.1	<0.05	0.073	<0.005	94	9.1	370	107	2.86	0.5410	8.35	6.12
Average		152	0.73	<0.1	<0.05	0.033	<0.005	129	4.7 ^k	470	118	5.38 ^b	0.6842	8.35	7.57
2006	Epilimnion														
DATE	DEPTH	ALK	TKN	NH ₃ -N	NO ₂ +NO ₃ -N	TP	SRP	Cl ⁻	TSS	TS	TVS	SECCHI	COND	pH	DO
11-May	3	121	0.76	<0.1	<0.05	0.027	0.006	134	1.1	436	121	9.10 ^a	0.7416	8.95	10.66
15-Jun	3	114	0.77	<0.1	<0.05	0.057	0.014	131	2.7	445	138	7.87	0.7056	9.35	10.61
13-Jul	3	111	1.43	<0.1	<0.05	0.066	<0.005	134	6.8	465	162	4.49	0.6947	9.51	9.00
10-Aug	3	134	1.69	<0.1	<0.05	0.099	<0.005	135	11.0	466	147	2.33	0.7335	8.82	7.23
14-Sep	3	150	2.15	0.161	<0.05	0.145	<0.005	129	12.0	448	140	1.54	0.7224	8.30	5.65
Average		126	1.36	0.161 ^k	<0.05	0.079	0.010 ^k	133	6.7	452	142	5.07 ^b	0.7196	8.99	8.63
2005	Epilimnion														
DATE	DEPTH	ALK	TKN	NH ₃ -N	NO ₃ -N [*]	TP	SRP	Cl ⁻	TSS	TS	TVS	SECCHI	COND	pH	DO
11-May	3	136	0.67	<0.1	<0.05	0.026	<0.005	111	1.4	429	99	0 ^c	0.7135	8.05	9.34
15-Jun	3	159	0.92	0.130	<0.05	0.056	0.015	118	3.1	467	122	5.12	0.7788	7.63	4.85
13-Jul	3	155	0.80	<0.1	<0.05	0.045	0.011	123	3.6	483	140	4.86	0.7855	7.71	5.89
10-Aug	3	152	1.08	<0.1	<0.05	0.063	<0.005	128	5.2	468	135	3.61	0.7519	8.95	8.07
14-Sep	3	147	1.11	<0.1	<0.05	0.066	<0.005	132	6.8	478	149	2.76	0.7435	8.77	6.53
Average		150	0.92	0.130 ^k	<0.05	0.051	0.013 ^k	122	4.0	465	129	4.09 ^d	0.7546	8.22	6.94

Glossary

ALK = Alkalinity, mg/L CaCO₃
 TKN = Total Kjeldahl nitrogen, mg/L
 NH₃-N = Ammonia nitrogen, mg/L
 NO₂+NO₃-N = Nitrate + Nitrite nitrogen, mg/L
 NO₃-N = Nitrate + Nitrite nitrogen, mg/L
 TP = Total phosphorus, mg/L
 SRP = Soluble reactive phosphorus, mg/L
 Cl⁻ = Chloride, mg/L

TDS = Total dissolved solids, mg/L
 TSS = Total suspended solids, mg/L
 TS = Total solids, mg/L
 TVS = Total volatile solids, mg/L
 SECCHI = Secchi disk depth, ft.
 COND = Conductivity, milliSiemens/cm
 DO = Dissolved oxygen, mg/L

a = Secchi depth was obstructed by the bottom
 b = Secchi disk was on the bottom at least one month and therefore the average could have been deeper
 c = Secchi depth was obstructed by plants
 d = Secchi disk depth average does not include data from May because Secchi disk was obstructed by plants
 k = Denotes that the actual value is known to be less than the value presented.
 NA= Not applicable
 * = Prior to 2006 only Nitrate - nitrogen was analyzed

Table 1. Continued.

2008		Hypolimnion													
DATE	DEPTH	ALK	TKN	NH ₃ -N	NO ₂ +NO ₃ -N	TP	SRP	Cl ⁻	TSS	TS	TVS	SECCHI	COND	pH	DO
14-May	6	128	0.65	<0.1	<0.05	0.035	<0.005	126	2.6	415	97	NA	0.7175	8.97	10.1
11-Jun	6	132	0.68	<0.1	<0.05	0.042	0.007	118	2.5	409	102	NA	0.6979	8.58	7.58
Average		130	0.67	<0.1	<0.05	0.039	0.007 ^k	122	2.6	412	100	NA	0.7077	8.78	8.84
2007		Hypolimnion													
DATE	DEPTH	ALK	TKN	NH ₃ -N	NO ₂ +NO ₃ -N	TP	SRP	Cl ⁻	TSS	TS	TVS	SECCHI	COND	pH	DO
08-May	6	167	0.57	<0.1	<0.05	0.027	<0.005	133	3.6	514	119	NA	0.8609	8.06	8.22
12-Jun	6	159	0.61	<0.1	<0.05	0.014	<0.005	139	1.2	523	138	NA	0.8451	8.39	9.20
10-Jul	6	152	0.68	<0.1	<0.05	0.023	<0.005	143	3.2	504	129	NA	0.8470	8.29	6.37
07-Aug	6	133	0.70	<0.1	<0.05	0.037	<0.005	140	3.6	485	143	NA	0.7851	8.49	6.38
11-Sep	6	151	0.91	<0.1	<0.05	0.063	<0.005	94	8.5	366	79	NA	0.6273	8.17	6.05
Average		152	0.69	<0.1	<0.05	0.033	<0.005	130	4.0	478	122	NA	0.7931	8.28	7.24
2006		Hypolimnion													
DATE	DEPTH	ALK	TKN	NH ₃ -N	NO ₂ +NO ₃ -N	TP	SRP	Cl ⁻	TSS	TS	TVS	SECCHI	COND	pH	DO
11-May	6	121	0.68	<0.1	<0.05	0.029	0.006	135	1.0	433	116	NA	0.7403	8.97	10.54
15-Jun	6	114	0.75	<0.1	<0.05	0.057	0.015	132	2.3	456	149	NA	0.7058	9.38	10.80
13-Jul	6	111	1.40	<0.1	<0.05	0.063	<0.005	136	6.6	448	146	NA	0.6954	9.46	8.15
10-Aug	6	134	1.58	<0.1	<0.05	0.082	<0.005	135	9.9	441	124	NA	0.7335	8.74	6.51
14-Sep	6	151	1.97	0.171	<0.05	0.132	<0.005	127	11.0	443	131	NA	0.7229	8.22	4.97
Average		126	1.28	0.171 ^k	<0.05	0.073	0.011 ^k	133	6.2	444	133	NA	0.7196	8.95	8.19
2005		Hypolimnion													
DATE	DEPTH	ALK	TKN	NH ₃ -N	NO ₃ -N*	TP	SRP	Cl ⁻	TSS	TS	TVS	SECCHI	COND	pH	DO
11-May	6	134	0.76	<0.1	<0.05	0.048	<0.005	112	18.6	441	109	NA	0.7005	8.14	11.80
15-Jun	6	159	0.89	0.138	<0.05	0.052	0.012	119	3.3	471	116	NA	0.7788	7.61	4.82
13-Jul	6	155	0.79	<0.1	<0.05	0.052	0.009	122	3.6	483	134	NA	0.7854	7.70	5.87
10-Aug	6	151	1.01	<0.1	<0.05	0.054	<0.005	128	4.4	460	126	NA	0.7535	8.84	7.08
Average		150	0.86	0.138 ^k	<0.05	0.052	0.011 ^k	120	7.5	464	121	NA	0.7546	8.07	7.39

Glossary

ALK = Alkalinity, mg/L CaCO ₃	TDS = Total dissolved solids, mg/L
TKN = Total Kjeldahl nitrogen, mg/L	TSS = Total suspended solids, mg/L
NH ₃ -N = Ammonia nitrogen, mg/L	TS = Total solids, mg/L
NO ₂ +NO ₃ -N = Nitrate + Nitrite nitrogen, mg/L	TVS = Total volatile solids, mg/L
NO ₃ -N = Nitrate + Nitrite nitrogen, mg/L	SECCHI = Secchi disk depth, ft.
TP = Total phosphorus, mg/L	COND = Conductivity, milliSiemens/cm
SRP = Soluble reactive phosphorus, mg/L	DO = Dissolved oxygen, mg/L
Cl ⁻ = Chloride, mg/L	

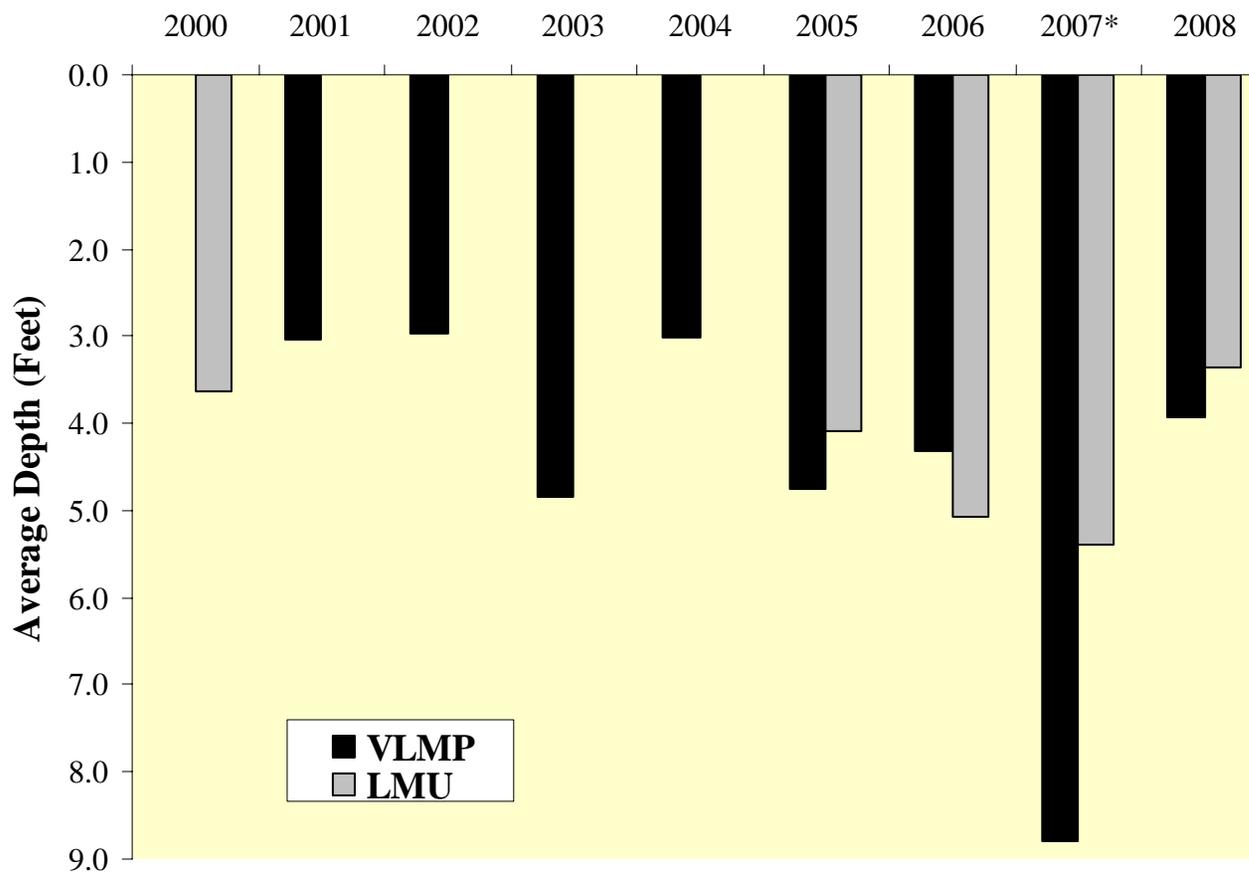
k = Denotes that the actual value is known to be less than the value presented.

NA= Not applicable

* = Prior to 2006 only Nitrate - nitrogen was analyzed

Figure 2. Yearly Secchi depth averages from VLMP and LCHD records for Countryside Lake.

* During 2007 VLMP data was only collected from May – June.



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Figure 3. Total suspended solid (TSS) concentrations vs. Secchi depth for Countryside Lake, 2008.

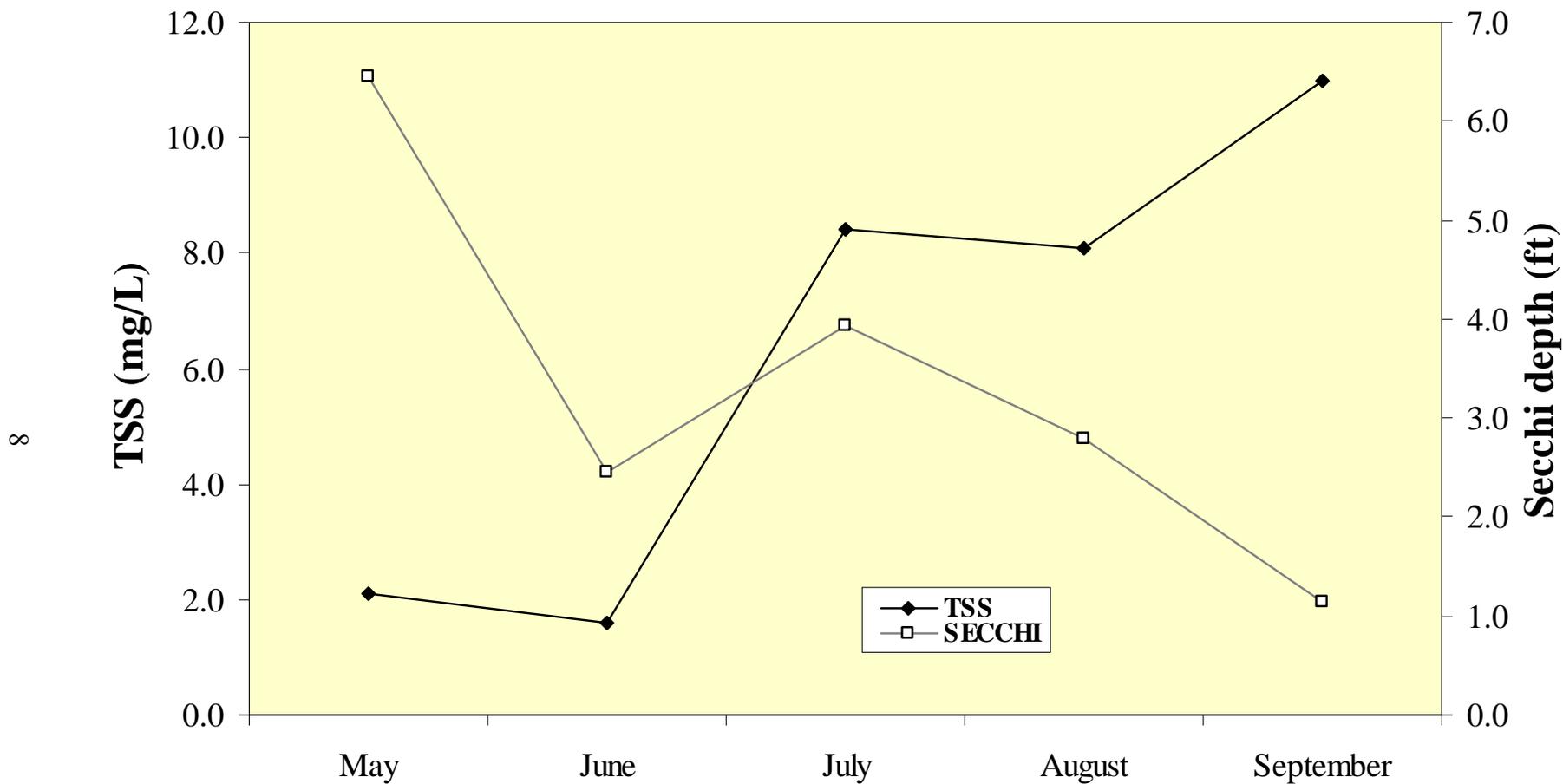


Table 2. Lake County average TSI phosphorous (TSIp) ranking 2000-2008.

RANK	LAKE NAME	TP AVE	TSIp
1	Lake Carina	0.0100	37.35
2	Sterling Lake	0.0100	37.35
3	Independence Grove	0.0135	39.24
4	Lake Zurich	0.0130	41.14
5	Sand Pond (IDNR)	0.0165	41.36
6	West Loon Lake	0.0140	42.21
7	Windward Lake	0.0158	43.95
8	Bangs Lake	0.0170	45.00
9	Pulaski Pond	0.0180	45.83
10	Timber Lake	0.0180	45.83
11	Fourth Lake	0.0182	45.99
12	Lake Kathryn	0.0200	47.35
13	Lake of the Hollow	0.0200	47.35
14	Banana Pond	0.0202	47.49
15	Lake Minear	0.0204	47.63
16	Cedar Lake	0.0220	48.72
17	Cross Lake	0.0220	48.72
18	Sun Lake	0.0220	48.72
19	Dog Pond	0.0222	48.85
20	Stone Quarry Lake	0.0230	49.36
21	Deep Lake	0.0234	49.61
22	Druce Lake	0.0244	50.22
23	Little Silver	0.0250	50.57
24	Round Lake	0.0254	50.80
25	Lake Leo	0.0256	50.91
26	Cranberry Lake	0.0270	51.68
27	Dugdale Lake	0.0274	51.89
28	Peterson Pond	0.0274	51.89
29	Lake Miltmore	0.0276	51.99
30	Third Lake	0.0280	52.20
31	Lake Fairfield	0.0296	53.00
32	Gray's Lake	0.0302	53.29
33	Highland Lake	0.0302	53.29
34	Hook Lake	0.0302	53.29
35	Lake Catherine (Site 1)	0.0308	53.57
36	Lambs Farm Lake	0.0312	53.76
37	Old School Lake	0.0312	53.76
38	Sand Lake	0.0316	53.94
39	Sullivan Lake	0.0320	54.13
40	Lake Linden	0.0326	54.39
41	Gages Lake	0.0338	54.92
42	Honey Lake	0.0340	55.00
43	Hendrick Lake	0.0344	55.17
44	Diamond Lake	0.0372	56.30
45	Channel Lake (Site 1)	0.0380	56.60
46	Ames Pit	0.0390	56.98

Table 2. Continued.

RANK	LAKE NAME	TP AVE	TSIp
47	White Lake	0.0408	57.63
48	Potomac Lake	0.0424	58.18
49	Duck Lake	0.0426	58.25
50	Old Oak Lake	0.0428	58.32
51	Deer Lake	0.0434	58.52
52	Schreiber Lake	0.0434	58.52
53	Nielsen Pond	0.0448	58.98
54	Turner Lake	0.0458	59.30
55	Seven Acre Lake	0.0460	59.36
56	Willow Lake	0.0464	59.48
57	Lucky Lake	0.0476	59.85
58	Davis Lake	0.0476	59.85
59	East Meadow Lake	0.0478	59.91
60	East Loon Lake	0.0490	60.27
61	College Trail Lake	0.0496	60.45
62	Lake Lakeland Estates	0.0524	61.24
63	Butler Lake	0.0528	61.35
64	West Meadow Lake	0.0530	61.40
65	Heron Pond	0.0545	61.80
66	Little Bear Lake	0.0550	61.94
67	Lucy Lake	0.0552	61.99
68	Lake Christa	0.0576	62.60
69	Lake Charles	0.0580	62.70
70	Crooked Lake	0.0608	63.38
71	Waterford Lake	0.0610	63.43
72	Lake Naomi	0.0616	63.57
73	Lake Tranquility S1	0.0618	63.62
74	Wooster Lake	0.0620	63.66
75	Countryside Lake	0.0620	63.66
76	Werhane Lake	0.0630	63.89
77	Liberty Lake	0.0632	63.94
78	Countryside Glen Lake	0.0642	64.17
79	Lake Fairview	0.0648	64.30
80	Leisure Lake	0.0648	64.30
81	Tower Lake	0.0662	64.61
82	St. Mary's Lake	0.0666	64.70
83	Mary Lee Lake	0.0682	65.04
84	Hastings Lake	0.0684	65.08
85	Spring Lake	0.0726	65.94
86	ADID 203	0.0730	66.02
87	Bluff Lake	0.0734	66.10
88	Harvey Lake	0.0766	66.71
89	Broberg Marsh	0.0782	67.01
90	Sylvan Lake	0.0794	67.23
91	Big Bear Lake	0.0806	67.45
92	Petite Lake	0.0834	67.94

Table 2. Continued.

RANK	LAKE NAME	TP AVE	TSIp
93	Timber Lake (South)	0.0848	68.18
94	Lake Marie (Site 1)	0.0850	68.21
95	North Churchill Lake	0.0872	68.58
96	Grand Avenue Marsh	0.0874	68.61
97	Grandwood Park, Site II, Outflow	0.0876	68.65
98	North Tower Lake	0.0878	68.68
99	South Churchill Lake	0.0896	68.97
100	Rivershire Pond 2	0.0900	69.04
101	McGreal Lake	0.0914	69.26
102	International Mine and Chemical Lake	0.0948	69.79
103	Eagle Lake (Site I)	0.0950	69.82
104	Valley Lake	0.0950	69.82
105	Dunns Lake	0.0952	69.85
106	Fish Lake	0.0956	69.91
107	Lochanora Lake	0.0960	69.97
108	Owens Lake	0.0978	70.23
109	Woodland Lake	0.0986	70.35
110	Island Lake	0.0990	70.41
111	McDonald Lake 1	0.0996	70.50
112	Longview Meadow Lake	0.1024	70.90
113	Lake Barrington	0.1053	71.31
114	Redwing Slough, Site II, Outflow	0.1072	71.56
115	Lake Forest Pond	0.1074	71.59
116	Bittersweet Golf Course #13	0.1096	71.88
117	Fox Lake (Site 1)	0.1098	71.90
118	Osprey Lake	0.1108	72.04
119	Bresen Lake	0.1126	72.27
120	Round Lake Marsh North	0.1126	72.27
121	Deer Lake Meadow Lake	0.1158	72.67
122	Long Lake	0.1170	72.82
123	Taylor Lake	0.1184	72.99
124	Columbus Park Lake	0.1226	73.49
125	Nippersink Lake (Site 1)	0.1240	73.66
126	Echo Lake	0.1250	73.77
127	Grass Lake (Site 1)	0.1288	74.21
128	Lake Holloway	0.1322	74.58
129	Lakewood Marsh	0.1330	74.67
130	Summerhill Estates Lake	0.1384	75.24
131	Redhead Lake	0.1412	75.53
132	Forest Lake	0.1422	75.63
133	Antioch Lake	0.1448	75.89
134	Slocum Lake	0.1496	76.36
135	Drummond Lake	0.1510	76.50
136	Pond-a-Rudy	0.1514	76.54
137	Lake Matthews	0.1516	76.56
138	Buffalo Creek Reservoir	0.1550	76.88

Table 2. Continued.

RANK	LAKE NAME	TP AVE	TSIp
139	Pistakee Lake (Site 1)	0.1592	77.26
140	Grassy Lake	0.1610	77.42
141	Salem Lake	0.1650	77.78
142	Half Day Pit	0.1690	78.12
143	Lake Eleanor Site II, Outflow	0.1812	79.13
144	Lake Farmington	0.1848	79.41
145	Lake Louise	0.1850	79.43
146	ADID 127	0.1886	79.71
147	Dog Bone Lake	0.1990	80.48
148	Redwing Marsh	0.2072	81.06
149	Stockholm Lake	0.2082	81.13
150	Bishop Lake	0.2156	81.63
151	Hidden Lake	0.2236	82.16
152	Fischer Lake	0.2278	82.43
153	Lake Napa Suwe (Outlet)	0.2304	82.59
154	Patski Pond (outlet)	0.2512	83.84
155	Oak Hills Lake	0.2792	85.36
156	Loch Lomond	0.2954	86.18
157	McDonald Lake 2	0.3254	87.57
158	Fairfield Marsh	0.3264	87.61
159	ADID 182	0.3280	87.69
160	Slough Lake	0.4134	91.02
161	Flint Lake Outlet	0.4996	93.75
162	Rasmussen Lake	0.5025	93.84
163	Albert Lake, Site II, outflow	1.1894	106.26

SUMMARY OF AQUATIC MACROPHYTES

Aquatic plant (macrophyte) surveys were conducted in May and August of 2008. Sampling sites were based on a grid system created by mapping software (ArcGIS), with each site located 60 meters apart. On Countryside Lake there were 159 sites sampled in May (Figure 4) and August (Figure 5). Plants were found at 149 sites in May and 82 sites in August (Table 3a, b). Overall, a total of three plant species and one macro-algae (*Chara* spp.) were found (Table 4). Only *Chara* spp. and Curlyleaf Pondweed were found in May, at 38% and 79% of the sampling sites, respectively. In August, *Chara* spp. was found at 47% of the sampling sites and Curlyleaf Pondweed at 2% of the sampling sites. Southern Naiad (8%) and Duckweed (2%) were other species found. These aquatic plant densities did not affect the recreational use of Countryside Lake. Diversity was similar to 2005, but down from 2007 and 2006 when four and five aquatic plant species, respectively, and *Chara* spp. were found. In 2000 12 aquatic plant species and one macro-algae were present. Countryside Lake was dominated by Curlyleaf Pondweed in May and *Chara* spp. in August during all sampling years. During 2008, the 1% light level reached the bottom from May through August. In May, plants were found down to 9.3 feet and in August they were found down to 9.0 feet.

CLA treated the aquatic plants with fluridone (Sonar AS™) at a concentration of 10 ppb on April 21, 2008 (5 gallons) and a bump up to hold concentration on May 6, 2008 (1.5 gallons). It was also treated with copper sulfate (a season total of 1145 pounds) to control algae from the end of April through early October. This was a decrease in the amount of copper sulfate applied during 2007 (1330+ pounds) and 2006 (2005 pounds). Even though there was a decrease, it may not be necessary to apply this much copper sulfate. Since LMU has been monitoring the lake, there has been a decrease in plant diversity, plant density, and water clarity and an increase in TSS and TP. Aquatic plants help to stabilize sediment and utilize phosphorous. Therefore, the current aquatic plant management plan should be evaluated for possible changes. This could save money and have less potential impact to the environment.

Floristic Quality Index (FQI) is a rapid assessment tool designed to evaluate the closeness of the flora of an area to that of undisturbed conditions. It can be used to: 1) identify natural areas, 2) compare the quality of different sites or different locations within a single site, 3) monitor long-term floristic trends, and 4) monitor habitat restoration efforts. A high FQI number indicates that there is a large number of sensitive, high quality plant species present in the lake. Non-native species were also included in the FQI calculations for Lake County lakes. The average FQI for 2000-2008 Lake County lakes was 13.6 (Table 5). Countryside Lake had a FQI of 8.7 in 2008. This was down from 2007 and 2006 when the FQIs were 10.5 and 12.5 and up from 2005 when the FQI was 5.8.

Figure 4. Aquatic plant sampling grid illustrating plant density on Countryside Lake, May 2008.

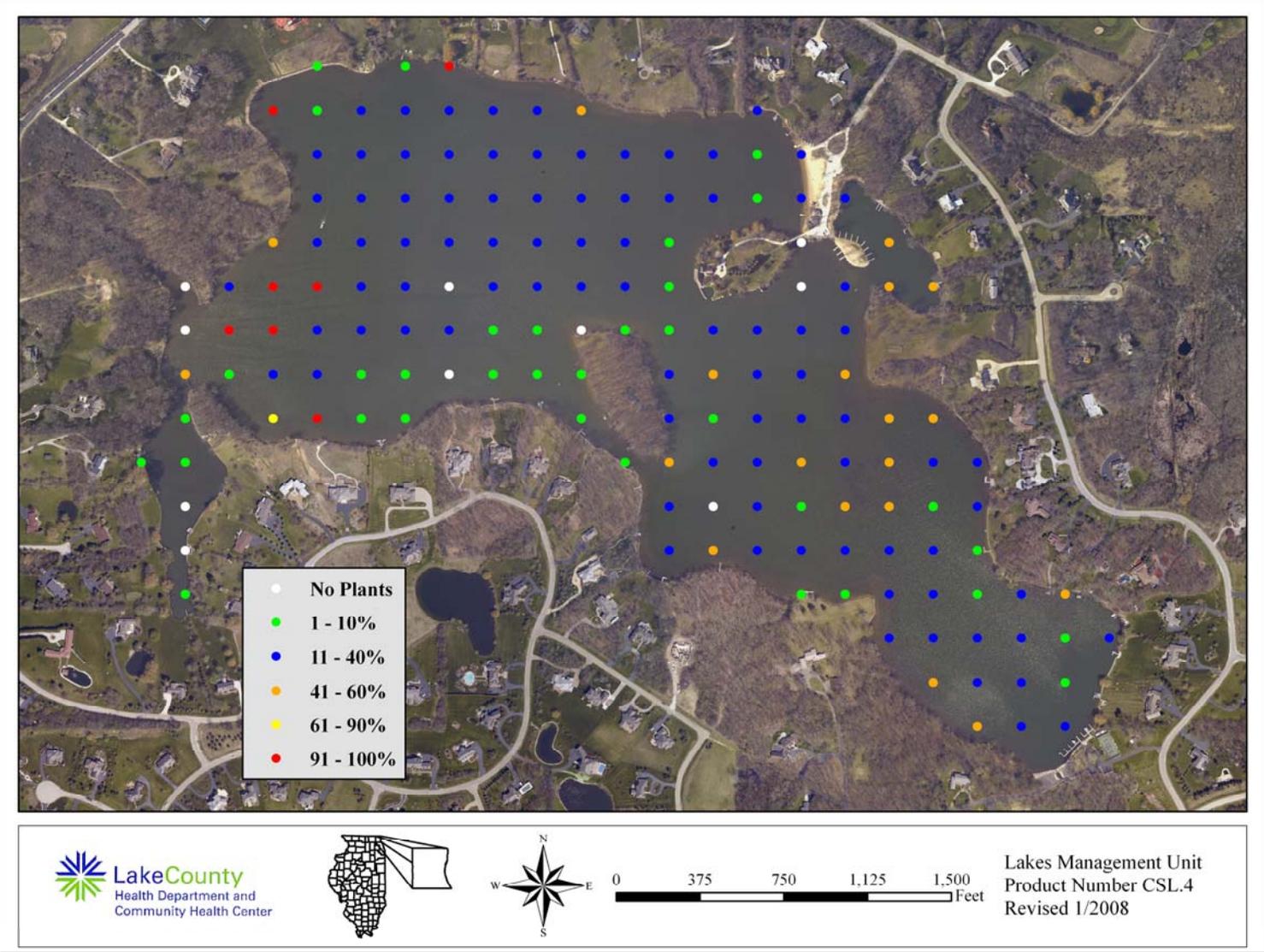
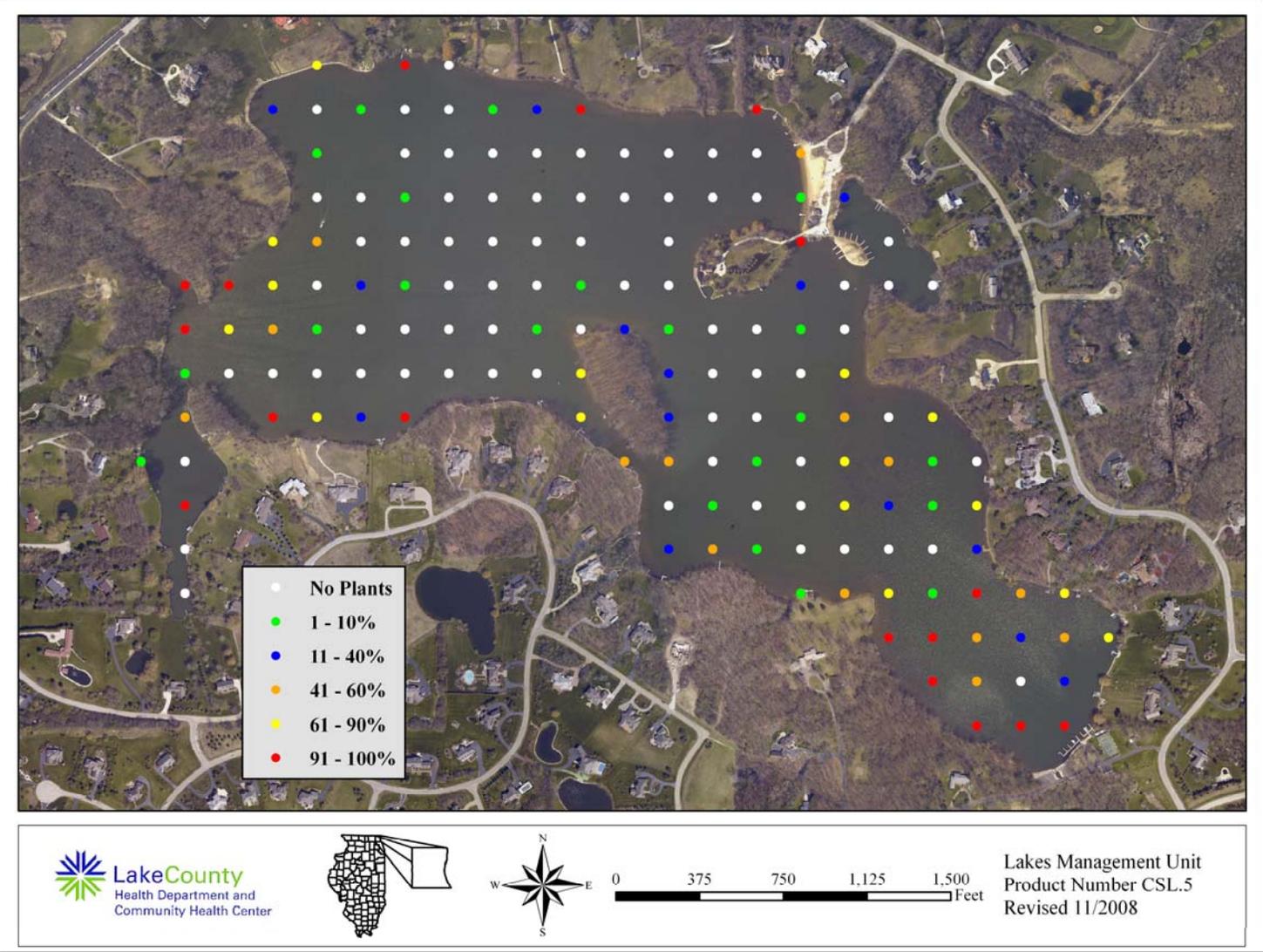


Figure 5. Aquatic plant sampling grid illustrating plant density on Countryside Lake, August 2008.



**Table 3a. Aquatic plant species found at the 159 sampling sites on Countryside Lake, 2008.
Maximum depth that plants were found was feet 9.3.**

May			August				
Plant Density	Chara	Curlyleaf Pondweed	Plant Density	Chara	Curlyleaf Pondweed	Duckweed	Southern Naiad
Absent	103	34	Absent	85	156	156	147
Present	21	40	Present	19	3	3	5
Common	18	75	Common	15	0	0	1
Abundant	12	7	Abundant	12	0	0	3
Dominant	5	3	Dominant	28	0	0	3
% Plant Occurrence	35.2	78.6	% Plant Occurrence	46.5	1.9	1.9	7.5

Table 3b. Distribution of rake density across all sampled sites.

May			August		
Rake Density (Coverage)	# of Sites	%	Rake Density (Coverage)	# of Sites	%
No plants	10	6.3	No plants	77	48.4
>0 to 10%	34	21.4	>0 to 10%	21	13.2
>10 to 40%	88	55.3	>10 to 40%	15	9.4
>40 to 60%	19	11.9	>40 to 60%	14	8.8
>60 to 90%	1	0.6	>60 to 90%	15	9.4
>90%	7	4.4	>90%	17	10.7
Total Sites with Plants	149	93.7	Total Sites with Plants	82	51.6
Total # of Sites	159	100.0	Total # of Sites	159	100.0

Table 4. Aquatic plant species found in Countryside Lake in 2008.

Chara (Macro algae)
Curlyleaf Pondweed[^]
Duckweed
Southern Naiad

Chara spp.
Potamogeton crispus
Lemna sp.
Najas guadalupensis

[^] **Exotic**

Table 5. Floristic quality index (FQI) of lakes in Lake County, calculated with exotic species (w/Adventives) and with native species only (native).

RANK	LAKE NAME	FQI (w/A)	FQI (native)
1	Cedar Lake	36.3	38.4
2	East Loon Lake	30.6	32.7
3	Cranberry Lake	30.1	31.6
4	Deep Lake	29.7	31.2
5	Little Silver	29.6	31.6
6	Round Lake Marsh North	29.1	29.9
7	Deer Lake	28.2	29.7
8	Sullivan Lake	28.2	29.7
9	Schreiber Lake	26.8	27.6
10	Bangs Lake	25.7	27.4
11	West Loon Lake	25.7	27.3
12	Cross Lake	25.2	27.8
13	Independence Grove	24.6	27.5
14	Sterling Lake	24.5	26.9
15	Lake Zurich	24.3	27.1
16	Sun Lake	24.3	26.1
17	Lake of the Hollow	23.8	26.2
18	Lakewood Marsh	23.8	24.7
19	Round Lake	23.5	25.9
20	Honey Lake	23.3	25.1
21	Fourth Lake	23.0	24.8
22	Druce Lake	22.8	25.2
23	Countryside Glen Lake	21.9	22.8
24	Butler Lake	21.4	23.1
25	Duck Lake	21.1	22.9
26	Timber Lake (North)	20.8	22.8
27	Broberg Marsh	20.5	21.4
28	Davis Lake	20.5	21.4
29	ADID 203	20.5	20.5
30	McGreal Lake	20.2	22.1
31	Lake Kathryn	19.6	20.7
32	Fish Lake	19.3	21.2
33	Owens Lake	19.3	20.2
34	Redhead Lake	19.3	21.2
35	Turner Lake	18.6	21.2
36	Wooster Lake	18.5	20.2
37	Salem Lake	18.5	20.2
38	Lake Miltmore	18.4	20.3
39	Hendrick Lake	17.7	17.7
40	Summerhill Estates Lake	17.1	18.0
41	Seven Acre Lake	17.0	15.5
42	Gray's Lake	16.9	19.8
43	Lake Barrington	16.7	17.7
44	Bresen Lake	16.6	17.8

Table 5. Continued.

Rank	LAKE NAME	FQI (w/A)	FQI (native)
45	Diamond Lake	16.3	17.4
46	Lake Napa Suwe	16.3	17.4
47	Windward Lake	16.3	17.6
48	Dog Bone Lake	15.7	15.7
49	Redwing Slough	15.6	16.6
50	Osprey Lake	15.5	17.3
51	Lake Fairview	15.2	16.3
52	Heron Pond	15.1	15.1
53	Lake Tranquility (S1)	15.0	17.0
54	North Churchill Lake	15.0	15.0
55	Dog Training Pond	14.7	15.9
56	Island Lake	14.7	16.6
57	Highland Lake	14.5	16.7
58	Grand Avenue Marsh	14.3	16.3
59	Taylor Lake	14.3	16.3
60	Dugdale Lake	14.0	15.1
61	Eagle Lake (S1)	14.0	15.1
62	Longview Meadow Lake	13.9	13.9
63	Ames Pit	13.4	15.5
64	Bishop Lake	13.4	15.0
65	Hook Lake	13.4	15.5
66	Long Lake	13.1	15.1
67	Buffalo Creek Reservoir	13.1	14.3
68	Mary Lee Lake	13.1	15.1
69	McDonald Lake 2	13.1	14.3
70	Old School Lake	13.1	15.1
71	Dunn's Lake	12.7	13.9
72	Old Oak Lake	12.7	14.7
73	Timber Lake (South)	12.7	14.7
74	White Lake	12.7	14.7
75	Hastings Lake	12.5	14.8
76	Sand Lake	12.5	14.8
77	Stone Quarry Lake	12.5	12.5
78	Lake Carina	12.1	14.3
79	Lake Leo	12.1	14.3
80	Lambs Farm Lake	12.1	14.3
81	Pond-A-Rudy	12.1	12.1
82	Stockholm Lake	12.1	13.5
83	Grassy Lake	12	12
84	Lake Matthews	12.0	12.0
85	Flint Lake	11.8	13.0
86	Harvey Lake	11.8	13.0
87	Rivershire Pond 2	11.5	13.3
88	Antioch Lake	11.3	13.4
89	Lake Charles	11.3	13.4
90	Lake Linden	11.3	11.3

Table 5. Continued.

Rank	LAKE NAME	FQI (w/A)	FQI (native)
91	Lake Naomi	11.2	12.5
92	Pulaski Pond	11.2	12.5
93	Lake Minear	11.0	13.9
94	Redwing Marsh	11.0	11.0
95	Tower Lake	11.0	11.0
96	West Meadow Lake	11.0	11.0
97	Nielsen Pond	10.7	12.0
98	Lake Holloway	10.6	10.6
99	Third Lake	10.2	12.5
100	Crooked Lake	10.2	12.5
101	College Trail Lake	10.0	10.0
102	Lake Lakeland Estates	10.0	11.5
103	Valley Lake	9.9	9.9
104	Werhane Lake	9.8	12.0
105	Big Bear Lake	9.5	11.0
106	Little Bear Lake	9.5	11.0
107	Loch Lomond	9.4	12.1
108	Columbus Park Lake	9.2	9.2
109	Sylvan Lake	9.2	9.2
110	Lake Louise	9	10.4
111	Fischer Lake	9.0	11.0
112	Grandwood Park Lake	9.0	11.0
113	Lake Fairfield	9.0	10.4
114	McDonald Lake 1	8.9	10.0
115	Countryside Lake	8.7	10.6
116	East Meadow Lake	8.5	8.5
117	Lake Christa	8.5	9.8
118	Lake Farmington	8.5	9.8
119	Lucy Lake	8.5	9.8
120	South Churchill Lake	8.5	8.5
121	Bittersweet Golf Course #13	8.1	8.1
122	Woodland Lake	8.1	9.9
123	Albert Lake	7.5	8.7
124	Banana Pond	7.5	9.2
125	Fairfield Marsh	7.5	8.7
126	Lake Eleanor	7.5	8.7
127	Patski Pond	7.1	7.1
128	Rasmussen Lake	7.1	7.1
129	Slough Lake	7.1	7.1
130	Lucky Lake	7.0	7.0
131	Lake Forest Pond	6.9	8.5
132	Leisure Lake	6.4	9.0
133	Peterson Pond	6.0	8.5
134	Gages Lake	5.8	10.0
135	Slocum Lake	5.8	7.1
136	Deer Lake Meadow Lake	5.2	6.4

Table 5. Continued.

Rank	LAKE NAME	FQI (w/A)	FQI (native)
137	ADID 127	5.0	5.0
138	Drummond Lake	5.0	7.1
139	IMC Lake	5.0	7.1
140	Liberty Lake	5.0	5.0
141	Oak Hills Lake	5.0	5.0
142	Forest Lake	3.5	5.0
143	Sand Pond (IDNR)	3.5	5.0
144	Half Day Pit	2.9	5.0
145	Lochanora Lake	2.5	5.0
146	Echo Lake	0.0	0.0
147	Hidden Lake	0.0	0.0
148	North Tower Lake	0.0	0.0
149	Potomac Lake	0.0	0.0
150	St. Mary's Lake	0.0	0.0
151	Waterford Lake	0.0	0.0
152	Willow Lake	0.0	0.0
	<i>Mean</i>	13.6	14.9
	<i>Median</i>	12.5	14.3

**APPENDIX A. METHODS FOR FIELD DATA COLLECTION AND
LABORATORY ANALYSES**

Water Sampling and Laboratory Analyses

Two water samples were collected once a month from May through September. Sample locations were at the deepest point in the lake (see sample site map), three feet below the surface, and 3 feet above the bottom. Samples were collected with a horizontal Van Dorn water sampler. Approximately three liters of water were collected for each sample for all lab analyses. After collection, all samples were placed in a cooler with ice until delivered to the Lake County Health Department lab, where they were refrigerated. Analytical methods for the parameters are listed in Table A1. Except nitrate nitrogen, all methods are from the Eighteenth Edition of Standard Methods, (eds. American Public Health Association, American Water Works Association, and Water Pollution Control Federation, 1992). Methodology for nitrate nitrogen was taken from the 14th edition of Standard Methods. Dissolved oxygen, temperature, conductivity and pH were measured at the deep hole with a Hydrolab DataSonde® 4a. Photosynthetic Active Radiation (PAR) was recorded using a LI-COR® 192 Spherical Sensor attached to the Hydrolab DataSonde® 4a. Readings were taken at the surface and then every two feet until reaching the bottom.

Plant Sampling

In order to randomly sample each lake, mapping software (ArcMap 9.3) overlaid a grid pattern onto an aerial photo of Lake County and placed points 60 or 30 meters apart, depending on lake size. Plants were sampled using a garden rake fitted with hardware cloth. The hardware cloth surrounded the rake tines and is tapered two feet up the handle. A rope was tied to the end of the handle for retrieval. At designated sampling sites, the rake was tossed into the water, and using the attached rope, was dragged across the bottom, toward the boat. After pulling the rake into the boat, plant coverage was assessed for overall abundance. Then plants were individually identified and placed in categories based on coverage. Plants that were not found on the rake but were seen in the immediate vicinity of the boat at the time of sampling were also recorded. Plants difficult to identify in the field were placed in plastic bags and identified with plant keys after returning to the office. The depth of each sampling location was measured either by a hand-held depth meter, or by pushing the rake straight down and measuring the depth along the rope or rake handle. One-foot increments were marked along the rope and rake handle to aid in depth estimation.

Wildlife Assessment

Species of wildlife were noted during visits to each lake. When possible, wildlife was identified to species by sight or sound. However, due to time constraints, collection of quantitative information was not possible. Thus, all data should be considered anecdotal. Some of the species on the list may have only been seen once, or were spotted during their migration through the area.

Table A1. Analytical methods used for water quality parameters.

<i>Parameter</i>	<i>Method</i>
Temperature	Hydrolab DataSonde® 4a or YSI 6600 Sonde®
Dissolved oxygen	Hydrolab DataSonde® 4a or YSI 6600 Sonde®
Nitrate and Nitrite nitrogen	USEPA 353.2 rev. 2.0 EPA-600/R-93/100 Detection Limit = 0.05 mg/L
Ammonia nitrogen	SM 18 th ed. Electrode method, #4500 NH ₃ -F Detection Limit = 0.1 mg/L
Total Kjeldahl nitrogen	SM 18 th ed, 4500-N _{org} C Semi-Micro Kjeldahl, plus 4500 NH ₃ -F Detection Limit = 0.5 mg/L
pH	Hydrolab DataSonde® 4a, or YSI 6600 Sonde® Electrometric method
Total solids	SM 18 th ed, Method #2540B
Total suspended solids	SM 18 th ed, Method #2540D Detection Limit = 0.5 mg/L
Chloride	SM 18 th ed, Method #4500C1-D
Total volatile solids	SM 18 th ed, Method #2540E, from total solids
Alkalinity	SM 18 th ed, Method #2320B, potentiometric titration curve method
Conductivity	Hydrolab DataSonde® 4a or YSI 6600 Sonde®
Total phosphorus	SM 18 th ed, Methods #4500-P B 5 and #4500-P E Detection Limit = 0.01 mg/L
Soluble reactive phosphorus	SM 18 th ed, Methods #4500-P B 1 and #4500-P E Detection Limit = 0.005 mg/L
Clarity	Secchi disk
Color	Illinois EPA Volunteer Lake Monitoring Color Chart
Photosynthetic Active Radiation (PAR)	Hydrolab DataSonde® 4a or YSI 6600 Sonde®, LI-COR® 192 Spherical Sensor

**APPENDIX B. MULTI-PARAMETER DATA FOR COUNTRYSIDE LAKE
IN 2008.**

Countryside Lake 2008 Multiparameter data

		Text								Depth of Light Meter	% Light Transmission Average	Extinction Coefficient
Date	Time	Depth	Dep25	Temp	DO	DO%	SpCond	pH	PAR	feet		
MMDDYY	HHMMSS	feet	feet	øC	mg/l	Sat	mS/cm	Units	æE/s/mý	feet		0.41
51408	81315	0	0.30	15.53	10.10	105.2	0.7177	8.75	351	Surface		
51408	81351	1	1.09	15.54	10.11	105.3	0.7181	8.77	432	Surface	100%	
51408	81441	2	1.93	15.53	10.11	105.3	0.7175	8.81	306	0.18	71%	1.92
51408	81557	3	2.99	15.53	10.13	105.4	0.7177	8.87	148	1.24	34%	0.59
51408	81643	4	4.08	15.53	10.12	105.4	0.7175	8.9	101	2.33	23%	0.16
51408	81738	5	4.97	15.52	10.11	105.3	0.7175	8.93	87	3.22	20%	0.05
51408	81831	6	6.05	15.53	10.10	105.1	0.7175	8.97	75	4.3	17%	0.03
51408	81910	7	6.99	15.52	10.08	104.9	0.7171	9.00	57	5.24	13%	0.05
51408	82005	8	8.02	15.50	10.04	104.5	0.7173	9.01	42	6.27	10%	0.05

		Text								Depth of Light Meter	% Light Transmission Average	Extinction Coefficient
Date	Time	Depth	Dep25	Temp	DO	DO%	SpCond	pH	PAR	feet		
MMDDYY	HHMMSS	feet	feet	øC	mg/l	Sat	mS/cm	Units	æE/s/mý	feet		0.62
61108	75640	0	0.26	23.73	7.70	93.7	0.6981	8.51	3150	Surface		
61108	75819	1	1.14	23.76	7.65	93.1	0.6982	8.53	792	Surface	100%	
61108	75925	2	2.12	23.75	7.62	92.8	0.6982	8.57	204	0.37	26%	3.67
61108	80026	3	3.02	23.72	7.61	92.6	0.6980	8.58	111	1.27	14%	0.48
61108	80124	4	4.03	23.73	7.61	92.6	0.6982	8.60	138	2.28	17%	-0.10
61108	80323	5	4.99	23.7	7.61	92.5	0.6981	8.59	87	3.24	11%	0.14
61108	80436	6	6.12	23.66	7.58	92.1	0.6979	8.58	65	4.37	8%	0.07
61108	80617	7	7.05	23.65	7.56	91.8	0.6984	8.58	50	5.3	6%	0.05
61108	80717	8	7.91	23.59	7.35	89.2	0.6983	8.61	43	6.16	5%	0.02
61108	80842	9	9.15	22.48	3.26	38.8	0.6938	7.91	34	7.4	4%	0.03

Text										Depth of Light Meter	% Light Transmission Average	Extinction Coefficient
Date	Time	Depth	Dep25	Temp	DO	DO%	SpCond	pH	PAR	feet		
MMDDYY	HHMMSS	feet	feet	øC	mg/l	Sat	mS/cm	Units	æE/s/mý	feet		0.79
70908	73831	0	0.51	25.36	9.74	122.5	0.6887	9.08	3777	Surface		
70908	74037	1	1.19	25.38	9.96	125.3	0.6886	9.17	4135	Surface	100%	
70908	74120	2	2.01	25.37	9.95	125.2	0.6887	9.17	1340	0.26	32%	4.33
70908	74325	3	2.92	25.37	9.94	125.0	0.6889	9.18	899	1.17	22%	0.34
70908	74401	4	4.05	25.36	9.94	125.0	0.6889	9.20	356	2.3	9%	0.40
70908	74516	5	5.03	25.36	9.89	124.4	0.6891	9.20	230	3.28	6%	0.13
70908	74624	6	5.95	25.36	9.84	123.7	0.6891	9.21	148	4.2	4%	0.10
70908	74746	7	7.02	25.35	9.76	122.7	0.6894	9.20	85	5.27	2%	0.11
70908	74847	8	8.06	25.33	9.66	121.4	0.6896	9.19	53	6.31	1.3%	0.07
70908	75007	9	8.77	25.34	0.34	4.3	0.6894	8.10	34	7.02	0.8%	0.06

Text										Depth of Light Meter	% Light Transmission Average	Extinction Coefficient
Date	Time	Depth	Dep25	Temp	DO	DO%	SpCond	pH	PAR	feet		
MMDDYY	HHMMSS	feet	feet	øC	mg/l	Sat	mS/cm	Units	æE/s/mý	feet		0.83
81308	73750	0	0.47	24.44	8.74	108.4	0.6861	8.83	1141	Surface		
81308	73923	1	1.11	24.45	8.70	107.9	0.6855	8.83	918	Surface	100%	
81308	74014	2	1.98	24.46	8.68	107.7	0.6856	8.85	269	0.23	29%	5.34
81308	74112	3	2.99	24.45	8.68	107.7	0.6857	8.86	160	1.24	17%	0.42
81308	74156	4	4.03	24.45	8.64	107.2	0.6857	8.87	59	2.28	6%	0.44
81308	74247	5	5.06	24.45	8.62	106.9	0.6857	8.87	39	3.31	4%	0.13
81308	74336	6	6.01	24.45	8.60	106.7	0.6859	8.88	27	4.26	3%	0.09
81308	74445	7	7.11	24.43	7.74	96.0	0.6874	8.83	14	5.36	2%	0.12
81308	74649	8	8.08	24.3	2.21	27.3	0.6939	8.23	8	6.33	0.9%	0.09
81308	74727	9	8.78	24.22	1.14	14.1	0.6963	8.12	6	7.03	0.7%	0.04

Date MMDDYY	Time HHMMSS	Text								Depth of Light Meter feet	% Light Transmission Average	Extinction Coefficient 1.05
		Depth feet	Dep25 feet	Temp øC	DO mg/l	DO% Sat	SpCond mS/cm	pH Units	PAR æE/s/mý			
91008	75926	0	0.42	19.61	9.79	109.1	0.6816	8.76	3774	Surface		
91008	80015	1	0.99	19.63	9.88	110.1	0.6814	8.55	3689	Surface	100%	
91008	80111	2	1.95	19.57	9.78	108.8	0.6814	8.56	1045	0.2	28%	6.31
91008	80222	3	3.07	19.48	9.13	101.4	0.6816	8.52	289	1.32	8%	0.97
91008	80502	4	3.99	19.42	8.25	91.6	0.683	8.46	113	2.24	3%	0.42
91008	80613	5	5.09	19.37	7.73	85.7	0.684	8.42	54	3.34	1.5%	0.22
91008	80702	6	6.03	19.32	7.49	83.0	0.6845	8.39	26	4.28	0.7%	0.17
91008	80815	7	7.01	19.26	7.11	78.7	0.6849	8.36	11	5.26	0.3%	0.16
91008	80916	8	8.03	19.25	7.06	78.0	0.6849	8.34	7	6.28	0.2%	0.07
91008	81048	9	8.62	19.24	6.73	74.5	0.6850	8.32	5	6.87	0.1%	0.05

**APPENDIX C. INTERPRETING YOUR LAKE'S WATER QUALITY
DATA**

Lakes possess a unique set of physical and chemical characteristics that will change over time. These in-lake water quality characteristics, or parameters, are used to describe and measure the quality of lakes, and they relate to one another in very distinct ways. As a result, it is virtually impossible to change any one component in or around a lake without affecting several other components, and it is important to understand how these components are linked.

The following pages will discuss the different water quality parameters measured by Lake County Health Department staff, how these parameters relate to each other, and why the measurement of each parameter is important. The median values (the middle number of the data set, where half of the numbers have greater values, and half have lesser values) of data collected from Lake County lakes from 2000-2008 will be used in the following discussion.

Temperature and Dissolved Oxygen:

Water temperature fluctuations will occur in response to changes in air temperatures, and can have dramatic impacts on several parameters in the lake. In the spring and fall, lakes tend to have uniform, well-mixed conditions throughout the water column (surface to the lake bottom). However, during the summer, deeper lakes will separate into distinct water layers. As surface water temperatures increase with increasing air temperatures, a large density difference will form between the heated surface water and colder bottom water. Once this difference is large enough, these two water layers will separate and generally will not mix again until the fall. At this time the lake is thermally stratified. The warm upper water layer is called the *epilimnion*, while the cold bottom water layer is called the *hypolimnion*. In some shallow lakes, stratification and destratification can occur several times during the summer. If this occurs the lake is described as polymictic. Thermal stratification also occurs to a lesser extent during the winter, when warmer bottom water becomes separated from ice-forming water at the surface until mixing occurs during spring ice-out.

Monthly temperature profiles were established on each lake by measuring water temperature every foot (lakes \leq 15 feet deep) or every two feet (lakes $>$ 15 feet deep) from the lake surface to the lake bottom. These profiles are important in understanding the distribution of chemical/biological characteristics and because increasing water temperature and the establishment of thermal stratification have a direct impact on dissolved oxygen (DO) concentrations in the water column. If a lake is shallow and easily mixed by wind, the DO concentration is usually consistent throughout the water column. However, shallow lakes are typically dominated by either plants or algae, and increasing water temperatures during the summer speeds up the rates of photosynthesis and decomposition in surface waters. When many of the plants or algae die at the end of the growing season, their decomposition results in heavy oxygen consumption and can lead to an oxygen crash. In deeper, thermally stratified lakes, oxygen production is greatest in the top portion of the lake, where sunlight drives photosynthesis, and oxygen consumption is greatest near the bottom of a lake, where sunken organic matter accumulates and decomposes. The oxygen difference between the top and bottom water layers can be dramatic, with plenty of oxygen near the surface, but practically none near the bottom. The oxygen profiles measured during the water quality study can illustrate if

this is occurring. This is important because the absence of oxygen (anoxia) near the lake bottom can have adverse effects in eutrophic lakes resulting in the chemical release of phosphorus from lake sediment and the production of hydrogen sulfide (rotten egg smell) and other gases in the bottom waters. Low oxygen conditions in the upper water of a lake can also be problematic since all aquatic organisms need oxygen to live. Some oxygen may be present in the water, but at too low a concentration to sustain aquatic life. Oxygen is needed by all plants, virtually all algae and for many chemical reactions that are important in lake functioning. Most adult sport-fish such as largemouth bass and bluegill require at least 3 mg/L of DO in the water to survive. However, their offspring require at least 5 mg/L DO as they are more sensitive to DO stress. When DO concentrations drop below 3 mg/L, rough fish such as carp and green sunfish are favored and over time will become the dominant fish species.

External pollution in the form of oxygen-demanding organic matter (i.e., sewage, lawn clippings, soil from shoreline erosion, and agricultural runoff) or nutrients that stimulate the growth of excessive organic matter (i.e., algae and plants) can reduce average DO concentrations in the lake by increasing oxygen consumption. This can have a detrimental impact on the fish community, which may be squeezed into a very small volume of water as a result of high temperatures in the epilimnion and low DO levels in the hypolimnion.

Nutrients:

Phosphorus:

For most Lake County lakes, phosphorus is the nutrient that limits plant and algae growth. This means that any addition of phosphorus to a lake will typically result in algae blooms or high plant densities during the summer. The source of phosphorus to a lake can be external or internal (or both). External sources of phosphorus enter a lake through point (i.e., storm pipes and wastewater discharge) and non-point runoff (i.e., overland water flow). This runoff can pick up large amounts of phosphorus from agricultural fields, septic systems or impervious surfaces before it empties into the lake.

Internal sources of phosphorus originate within the lake and are typically linked to the lake sediment. In lakes with high oxygen levels (oxic), phosphorus can be released from the sediment through plants or sediment resuspension. Plants take up sediment-bound phosphorus through their roots, releasing it in small amounts to the water column throughout their life cycles, and in large amounts once they die and begin to decompose. Sediment resuspension can occur through biological or mechanical means. Bottom-feeding fish, such as common carp and black bullhead can release phosphorus by stirring up bottom sediment during feeding activities and can add phosphorus to a lake through their fecal matter. Sediment resuspension, and subsequent phosphorus release, can also occur via wind/wave action or through the use of artificial aerators, especially in shallow lakes. In lakes that thermally stratify, internal phosphorus release can occur from the sediment through chemical means. Once oxygen is depleted (anoxia) in the hypolimnion, chemical reactions occur in which phosphorus bound to iron complexes in the sediment becomes soluble and is released into the water column. This phosphorus is trapped in the hypolimnion and is unavailable to algae until fall turnover, and can cause algae blooms once

it moves into the sunlit surface water at that time. Accordingly, many of the lakes in Lake County are plagued by dense algae blooms and excessive, exotic plant coverage, which negatively affect DO levels, fish communities and water clarity.

Lakes with an average phosphorus concentration greater than 0.05 mg/L are considered nutrient rich. The median near surface total phosphorus (TP) concentration in Lake County lakes from 2000-2008 is 0.065 mg/L and ranged from a non-detectable minimum of <0.010 mg/L on five lakes to a maximum of 3.880 mg/L on Albert Lake. The median anoxic TP concentration in Lake County lakes from 2000-2008 was 0.181 mg/L and ranged from a minimum of 0.012 mg/L in Independence Grove Lake to a maximum of 3.880 mg/L in Taylor Lake.

The analysis of phosphorus also included soluble reactive phosphorus (SRP), a dissolved form of phosphorus that is readily available for plant and algae growth. SRP is not discussed in great detail in most of the water quality reports because SRP concentrations vary throughout the season depending on how plants and algae absorb and release it. It gives an indication of how much phosphorus is available for uptake, but, because it does not take all forms of phosphorus into account, it does not indicate how much phosphorus is truly present in the water column. TP is considered a better indicator of a lake's nutrient status because its concentrations remain more stable than soluble reactive phosphorus. However, elevated SRP levels are a strong indicator of nutrient problems in a lake.

Nitrogen:

Nitrogen is also an important nutrient for plant and algae growth. Sources of nitrogen to a lake vary widely, ranging from fertilizer and animal wastes, to human waste from sewage treatment plants or failing septic systems, to groundwater, air and rainfall. As a result, it is very difficult to control or reduce nitrogen inputs to a lake. Different forms of nitrogen are present in a lake under different oxic conditions. NH_4^+ (ammonium) is released from decomposing organic material under anoxic conditions and accumulates in the hypolimnion of thermally stratified lakes. If NH_4^+ comes into contact with oxygen, it is immediately converted to NO_2^- (nitrite) which is then oxidized to NO_3^- (nitrate). Therefore, in a thermally stratified lake, levels of NH_4^+ would only be elevated in the hypolimnion and levels of NO_3^- would only be elevated in the epilimnion. Both NH_4^+ and NO_3^- can be used as a nitrogen source by aquatic plants and algae. Total Kjeldahl nitrogen (TKN) is a measure of organic nitrogen plus ammonium. Adding the concentrations of TKN and nitrate together gives an indication of the amount of total nitrogen present in the water column. If inorganic nitrogen (NO_3^- , NO_2^- , NH_4^+) concentrations exceed 0.3 mg/L in spring, sufficient nitrogen is available to support summer algae blooms. However, low nitrogen levels do not guarantee limited algae growth the way low phosphorus levels do. Nitrogen gas in the air can dissolve in lake water and blue-green algae can "fix" atmospheric nitrogen, converting it into a usable form. Since other types of algae do not have the ability to do this, nuisance blue-green algae blooms are typically associated with lakes that are nitrogen limited (i.e., have low nitrogen levels).

The ratio of TKN plus nitrate nitrogen to total phosphorus (TN:TP) can indicate whether plant/algae growth in a lake is limited by nitrogen or phosphorus. Ratios of less than 10:1

suggest a system limited by nitrogen, while lakes with ratios greater than 20:1 are limited by phosphorus. It is important to know if a lake is limited by nitrogen or phosphorus because any addition of the limiting nutrient to the lake will, likely, result in algae blooms or an increase in plant density.

Solids:

Although several forms of solids (total solids, total suspended solids, total volatile solids, total dissolved solids) were measured each month by the Lakes Management Staff, total suspended solids (TSS) and total volatile solids (TVS) have the most impact on other variables and on the lake as a whole. TSS are particles of algae or sediment suspended in the water column. High TSS concentrations can result from algae blooms, sediment resuspension, and/or the inflow of turbid water, and are typically associated with low water clarity and high phosphorus concentrations in many lakes in Lake County. Low water clarity and high phosphorus concentrations, in turn, exacerbate the high TSS problem by leading to reduced plant density (which stabilize lake sediment) and increased occurrence of algae blooms. The median TSS value in epilimnetic waters in Lake County is 8.2 mg/L, ranging from below the 0.1 mg/L detection limit to 165 mg/L in Fairfield Marsh.

TVS represents the fraction of total solids that are organic in nature, such as algae cells, tiny pieces of plant material, and/or tiny animals (zooplankton) in the water column. High TVS values indicate that a large portion of the suspended solids may be made up of algae cells. This is important in determining possible sources of phosphorus to a lake. If much of the suspended material in the water column is determined to be resuspended sediment that is releasing phosphorus, this problem would be addressed differently than if the suspended material was made up of algae cells that were releasing phosphorus. The median TVS value was 132.8 mg/L, ranging from 34.0 mg/L in Pulaski Pond to 298.0 mg/L in Fairfield Marsh.

Total dissolved solids (TDS) are the amount of dissolved substances, such as salts or minerals, remaining in water after evaporation. These dissolved solids are discussed in further detail in the *Alkalinity* and *Conductivity* sections of this document. TDS concentrations were measured in Lake County lakes prior to 2004, but was discontinued due to the strong correlation of TDS to conductivity and chloride concentrations.

Water Clarity:

Water clarity (transparency) is not a chemical property of lake water, but is often an indicator of a lake's overall water quality. It is affected by a lake's water color, which is a reflection of the amount of total suspended solids and dissolved organic chemicals. Thus, transparency is a measure of particle concentration and is measured with a Secchi disk. Generally, the lower the clarity or Secchi depth, the poorer the water quality. A decrease in Secchi depth during the summer occurs as the result of an increase in suspended solids (algae or sediment) in the water column. Aquatic plants play an important role in the level of water clarity and can, in turn, be negatively affected by low clarity levels. Plants increase clarity by competing with algae for

resources and by stabilizing sediments to prevent sediment resuspension. A lake with a healthy plant community will almost always have higher water clarity than a lake without plants. Additionally, if the plants in a lake are removed (through herbicide treatment or the stocking of grass carp), the lake will probably become dominated by algae and Secchi depth will decrease. This makes it very difficult for plants to become re-established due to the lack of available sunlight and the lake will, most likely, remain turbid. Turbidity will be accelerated if the lake is very shallow and/or common carp are present. Shallow lakes are more susceptible to sediment resuspension through wind/wave action and are more likely to experience clarity problems if plants are not present to stabilize bottom sediment.

Common Carp are prolific fish that feed on invertebrates in the sediment. Their feeding activities stir up bottom sediment and can dramatically decrease water clarity in shallow lakes. As mentioned above, lakes with low water clarity are, generally, considered to have poor water quality. This is because the causes and effects of low clarity negatively impact the plant and fish communities, as well as the levels of phosphorus in a lake. The detrimental impacts of low Secchi depth to plants has already been discussed. Fish populations will suffer as water clarity decreases due to a lack of food and decreased ability to successfully hunt for prey. Bluegills are planktivorous fish and feed on invertebrates that inhabit aquatic plants. If low clarity results in the disappearance of plants, this food source will disappear too. Largemouth Bass and Northern Pike are piscivorous fish that feed on other fish and hunt by sight. As the water clarity decreases, these fish species find it more difficult to see and ambush prey and may decline in size as a result. This could eventually lead to an imbalance in the fish community. Phosphorus release from resuspended sediment could increase as water clarity and plant density decrease. This would then result in increased algae blooms, further reducing Secchi depth and aggravating all problems just discussed. The average Secchi depth for Lake County lakes is 3.12 feet. From 2000-2008, Fairfield Marsh and Patski Pond had the lowest Secchi depths (0.33 feet) and Bangs Lake had the highest (29.23 feet). As an example of the difference in Secchi depth based on plant coverage, South Churchill Lake, which had no plant coverage and large numbers of Common Carp in 2003 had an average Secchi depth of 0.73 feet (over four times lower than the county average), while Deep Lake, which had a diverse plant community and few carp had an average 2003 Secchi depth of 12.48 feet (almost four times higher than the county average).

Another measure of clarity is the use of a light meter. The light meter measures the amount of light at the surface of the lake and the amount of light at each depth in the water column. The amount of attenuation and absorption (decreases) of light by the water column are major factors controlling temperature and potential photosynthesis. Light intensity at the lake surface varies seasonally and with cloud cover, and decreases with depth. The deeper into the water column light penetrates, the deeper potential plant growth. The maximum depth at which algae and plants can grow underwater is usually at the depth where the amount of light available is reduced to 0.5%-1% of the amount of light available at the lake surface. This is called the euphotic (sunlit) zone. A general rule of thumb in Lake County is that the 1% light level is about 1 to 3 times the Secchi disk depth.

Alkalinity, Conductivity, Chloride, pH:

Alkalinity:

Alkalinity is the measurement of the amount of acid necessary to neutralize carbonate (CO_3^-) and bicarbonate (HCO_3^-) ions in the water, and represents the buffering capacity of a body of water. The alkalinity of lake water depends on the types of minerals in the surrounding soils and in the bedrock. It also depends on how often the lake water comes in contact with these minerals.

If a lake gets groundwater from aquifers containing limestone minerals such as calcium carbonate (CaCO_3) or dolomite (CaMgCO_3), alkalinity will be high. The median alkalinity in Lake County lakes (162 mg/L) is considered moderately hard according to the hardness classification scale of Brown, Skougstad and Fishman (1970). Because hard water (alkaline) lakes often have watersheds with fertile soils that add nutrients to the water, they usually produce more fish and aquatic plants than soft water lakes. Since the majority of Lake County lakes have a high alkalinity they are able to buffer the adverse effects of acid rain.

Conductivity and Chloride:

Conductivity is the inverse measure of the resistance of lake water to an electric flow. This means that the higher the conductivity, the more easily an electric current is able to flow through water. Since electric currents travel along ions in water, the more chemical ions or dissolved salts a body of water contains, the higher the conductivity will be. Accordingly, conductivity has been correlated to total dissolved solids and chloride ions. The amount of dissolved solids or conductivity of a lake is dependent on the lake and watershed geology, the size of the watershed flowing into the lake, the land uses within that watershed, and evaporation and bacterial activity. Many Lake County lakes have elevated conductivity levels in May, but not during any other month. This was because chloride, in the form of road salt, was washing into the lakes with spring rains, increasing conductivity. Most road salt is sodium chloride, calcium chloride, potassium chloride, magnesium chloride or ferrocyanide salts. Beginning in 2004, chloride concentrations are one of the parameters measured during the lake studies. Increased chloride concentrations may have a negative impact on aquatic organisms. Conductivity changes occur seasonally and with depth. For example, in stratified lakes the conductivity normally increases in the hypolimnion as bacterial decomposition converts organic materials to bicarbonate and carbonate ions depending on the pH of the water. These newly created ions increase the conductivity and total dissolved solids. Over the long term, conductivity is a good indicator of potential watershed or lake problems if an increasing trend is noted over a period of years. It is also important to know the conductivity of the water when fishery assessments are conducted, as electroshocking requires a high enough conductivity to properly stun the fish, but not too high as to cause injury or death.

pH:

pH is the measurement of hydrogen ion (H^+) activity in water. The pH of pure water is neutral at 7 and is considered acidic at levels below 7 and basic at levels above 7. Low pH levels of 4-5 are toxic to most aquatic life, while high pH levels (9-10) are not only toxic to aquatic life but may also result in the release of phosphorus from lake sediment. The presence of high plant densities can increase pH levels through photosynthesis, and lakes dominated by a large amount of plants or algae can experience large fluctuations in pH levels from day to night, depending on the rates of photosynthesis and respiration. Few, if any pH problems exist in Lake County lakes.

Typically, the flooded gravel mines in the county are more acidic than the glacial lakes as they have less biological activity, but do not usually drop below pH levels of 7. The median near surface pH value of Lake County lakes is 8.32, with a minimum of 7.06 in Deer Lake and a maximum of 10.28 in Round Lake Marsh North.

Eutrophication and Trophic State Index:

The word *eutrophication* comes from a Greek word meaning “well nourished.” This also describes the process in which a lake becomes enriched with nutrients. Over time, this is a lake’s natural aging process, as it slowly fills in with eroded materials from the surrounding watershed and with decaying plants. If no human impacts disturb the watershed or the lake, natural eutrophication can take thousands of years. However, human activities on a lake or in the watershed accelerate this process by resulting in rapid soil erosion and heavy phosphorus inputs. This accelerated aging process on a lake is referred to as cultural eutrophication. The term trophic state refers to the amount of nutrient enrichment within a lake system. *Oligotrophic* lakes are usually deep and clear with low nutrient levels, little plant growth and a limited fishery. *Mesotrophic* lakes are more biologically productive than oligotrophic lakes and have moderate nutrient levels and more plant growth. A lake labeled as *eutrophic* is high in nutrients and can support high plant densities and large fish populations. Water clarity is typically poorer than oligotrophic or mesotrophic lakes and dissolved oxygen problems may be present. A *hypereutrophic* lake has excessive nutrients, resulting in nuisance plant or algae growth. These lakes are often pea-soup green, with poor water clarity. Low dissolved oxygen may also be a problem, with fish kills occurring in shallow, hypereutrophic lakes more often than less enriched lakes. As a result, rough fish (tolerant of low dissolved oxygen levels) dominate the fish community of many hypereutrophic lakes. The categorization of a lake into a certain trophic state should not be viewed as a “good to bad” categorization, as most lake residents rate their lake based on desired usage. For example, a fisherman would consider a plant-dominated, clear lake to be desirable, while a water-skier might prefer a turbid lake devoid of plants. Most lakes in Lake County are eutrophic or hypereutrophic. This is primarily as a result of cultural eutrophication. However, due to the fertile soil in this area, many lakes (especially man-made) may have started out under eutrophic conditions and will never attain even mesotrophic conditions, regardless of any amount of money put into the management options. This is not an excuse to allow a lake to continue to deteriorate, but may serve as a reality check for lake owners attempting to create unrealistic conditions in their lakes.

The Trophic State Index (TSI) is an index which attaches a score to a lake based on its average

total phosphorus concentration, its average Secchi depth (water transparency) and/or its average chlorophyll *a* concentration (which represent algae biomass). It is based on the principle that as phosphorus levels increase, chlorophyll *a* concentrations increase and Secchi depth decreases. The higher the TSI score, the more nutrient-rich a lake is, and once a score is obtained, the lake can then be designated as oligotrophic, mesotrophic or eutrophic. Table 1 (below) illustrates the Trophic State Index using phosphorus concentration and Secchi depth.

Table 1. Trophic State Index (TSI).

Trophic State	TSI score	Total Phosphorus (mg/L)	Secchi Depth (feet)
Oligotrophic	<40	≤ 0.012	>13.12
Mesotrophic	$\geq 40 < 50$	$> 0.012 \leq 0.024$	$\geq 6.56 < 13.12$
Eutrophic	$\geq 50 < 70$	$> 0.024 \leq 0.096$	$\geq 1.64 < 6.56$
Hypereutrophic	≥ 70	> 0.096	< 1.64

**APPENDIX D. WATER QUALITY STATISTICS FOR ALL LAKE
COUNTY LAKES.**

2000 - 2008 Water Quality Parameters, Statistics Summary

	ALKoxic <=3ft00-2008		ALKanoxic 2000-2008	
Average	167		202	
Median	162		194	
Minimum	65	IMC	103	Heron Pond
Maximum	330	Flint Lake	470	Lake Marie
STD	42		50	
n =	802		243	

	Condoxic <=3ft00-2008		Condanoxic 2000-2008	
Average	0.8934		1.0312	
Median	0.8195		0.8695	
Minimum	0.2542	Broberg Marsh	0.3210	Lake Kathyrn
Maximum	6.8920	IMC	7.4080	IMC
STD	0.5250		0.7985	
n =	806		243	

	NO3-N, Nitrate+Nitrite,oxic <=3ft00-2008		NH3- Nanoxic 2000-2008	
Average	0.508		2.192	
Median	0.156		1.630	
Minimum	<0.05	*ND	<0.1	*ND
Maximum	9.670	South Churchill Lake	18.400	Taylor Lake
STD	1.073		2.343	
n =	807		243	

*ND = Many lakes had non-detects (74.1%)

*ND = 19.8% Non-detects from 28 different lakes

Only compare lakes with detectable concentrations to the statistics above
Beginning in 2006, Nitrate+Nitrite was measured.

	pHoxic <=3ft00-2008		pHanoxic 2000-2008	
Average	8.32		7.28	
Median	8.32		7.28	
Minimum	7.07	Bittersweet #13 Round Lake Marsh North	6.24	Banana Pond
Maximum	10.28		8.48	Heron Pond
STD	0.44		0.42	
n =	801		243	

	All Secchi 2000-2008	
Average	4.51	
Median	3.12	
Minimum	0.33	Fairfield Marsh, Patski Pon
Maximum	24.77	West Loon Lake
STD	3.78	
n =	749	



2000 - 2008 Water Quality Parameters, Statistics Summary (continued)

	TKNoxic <=3ft00-2008	
Average	1.450	
Median	1.200	
Minimum	<0.1	*ND
Maximum	10.300	Fairfield Marsh
STD	0.845	
n =	802	

*ND = 3.9% Non-detects from 15 different lakes

	TKNanoxic 2000-2008	
Average	2.973	
Median	2.330	
Minimum	<0.5	*ND
Maximum	21.000	Taylor Lake
STD	2.324	
n =	243	

*ND = 2.9% Non-detects from 4 different lakes

	TPoxic <=3ft00-2008	
Average	0.105	
Median	0.065	
Minimum	<0.01	*ND
Maximum	3.880	Albert Lake
STD	0.218	
n =	808	

*ND = 2.6% Non-detects from 9 different lakes

	TPanoxic 2000-2008	
Average	0.316	
Median	0.181	
Minimum	0.012	Independ. Grove
Maximum	3.800	Taylor Lake
STD	0.419	
n =	243	

	TSSall <=3ft00-2008	
Average	15.5	
Median	8.2	
Minimum	<0.1	*ND
Maximum	165.0	Fairfield Marsh
STD	20.3	
n =	813	

*ND = 1.5% Non-detects from 9 different lakes

	TVSoxic <=3ft00-2008	
Average	132.8	
Median	129.0	
Minimum	34.0	Pulaski Pond
Maximum	298.0	Fairfield Marsh
STD	39.8	
n =	757	

No 2002 IEPA Chain Lakes

	TDSoxic <=3ft00-2004	
Average	470	
Median	454	
Minimum	150	Lake Kathryn, White
Maximum	1340	IMC
STD	169	
n =	745	

No 2002 IEPA Chain Lakes.

	CLanoxic <=3ft00-2008	
Average	234	
Median	139	
Minimum	41	Timber Lake (N)
Maximum	2390	IMC
STD	364	
n =	125	

	CLoxic <=3ft00-2008	
Average	210	
Median	166	
Minimum	30	White Lake
Maximum	2760	IMC
STD	233	
n =	470	

Anoxic conditions are defined ≤ 1 mg/l D.O.
pH Units are equal to the -Log of [H] ion activity
Conductivity units are in MilliSiemens/cm
Secchi Disk depth units are in feet
All others are in mg/L

Minimums and maximums are based on data from all lakes from 2000-2008 (n=1351).

Average, median and STD are based on data from the most recent water quality sampling year for each lake.

LCHD Lakes Management Unit ~ 12/1/2008

APPENDIX E. GRANT PROGRAM OPPORTUNITES.

Table E1. Potential Grant Opportunities

Grant Program Name	Funding Source	Contact Information	Funding Focus				Cost Share
			Water Quality/ Wetland	Habitat	Erosion	Flooding	
Challenge Grant Program	USFWS	847-381-2253 or 309-793-5800		X	X		
Chicago Wilderness Small Grants	CW	312-346-8166 ext. 30					None
Partners in Conservation (formerly C2000)	IDNR	http://dnr.state.il.us/orep/c2000/		X			None
Conservation Reserve Program	NRCS	http://www.nrcs.usda.gov/programs/crp/		X			Land
Ecosystems Program	IDNR	http://dnr.state.il.us/orep/c2000/ecosystem/		X			None
Emergency Watershed Protection	NRCS	http://www.nrcs.usda.gov/programs/ewp/			X	X	None
Five Star Challenge	NFWF	http://www.nfwf.org/AM/Template.cfm		X			None
Illinois Flood Mitigation Assistance Program	IEMA	http://www.state.il.us/iema/construction.htm				X	None
Great Lakes Basin Program	GLBP	http://www.glc.org/basin/stateproj.html?st=il	X		X		None
Illinois Clean Energy Community Foundation	ICECF	http://www.illinoiscleanenergy.org/		X			
Illinois Clean Lakes Program	IEPA	http://www.epa.state.il.us/water/financial-assistance/index.html					None
Lake Education Assistance Program (LEAP)	IEPA	http://www.epa.state.il.us/water/conservation-2000/leap/index.html	X				\$500

CW = Chicago Wilderness
 ICECF = Illinois Clean Energy Community Foundation
 IEMA = Illinois Emergency Management Agency
 IEPA = Illinois Environmental Protection Agency
 IDNR = Illinois Department of Natural Resources
 IDOA = Illinois Department of Agriculture
 LCSCMC = Lake County Stormwater Management Commission
 LCSWCD = Lake County Soil and Water Conservation District
 NFWF = National Fish and Wildlife Foundation
 NRCS = Natural Resources Conservation Service
 USACE = United States Army Corps of Engineers
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Table E1. Continued

Grant Program Name	Funding Source	Contact Information	Funding Focus				Cost Share
			Water Quality/ Wetland	Habitat	Erosion	Flooding	
Northeast Illinois Wetland Conservation Account	USFWF	847-381-2253	X				
Partners for Fish and Wildlife	USFWS	http://ecos.fws.gov/partners/		X			> 50%
River Network's Watershed Assistance Grants Program	River Network	http://www.rivernetwork.org	X	X	X		na
Section 206: Aquatic Ecosystems Restoration	USACE	312-353-6400, 309-794-5590 or 314-331-8404		X			35%
Section 319: Non-Point Source Management Program	IEPA	http://www.epa.state.il.us/water/financial-assistance/non-point.html	X	X			>40%
Section 1135: Project Modifications for the Improvement of the Environment	USACE	312-353-6400, 309-794-5590 or 314-331-8404		X			25%
Stream Cleanup And Lakeshore Enhancement (SCALE)	IEPA	http://www.epa.state.il.us/water/watershed/scale.html	X	X			None
Streambank Stabilization & Restoration (SSRP)	IDOA/ LCSWCD	http://www.agr.state.il.us/Environment/conserv/ or call LCSWCD at (847) 223-1056		X	X		25%
Watershed Management Boards	LCSMC	http://www.co.lake.il.us/smc/projects/wmb/default.asp	X		X	X	50%
Wetlands Reserve Program	NRCS	http://www.nrcs.usda.gov/programs/wrp/	X	X			Land
Wildlife Habitat Incentive Program	NRCS	http://www.nrcs.usda.gov/programs/whip/		X			Land

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