

**2007 SUMMARY REPORT
of
Cranberry Lake**

Lake County, Illinois

Prepared by the

**LAKE COUNTY HEALTH DEPARTMENT
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LAKE FACTS

Lake Name:	Cranberry Lake
Historical Name:	None
Nearest Municipality:	Hainesville
Location:	T45N, R10E, Section 28
Elevation:	800.0 feet
Major Tributaries:	None
Watershed:	Fox River
Sub-watershed:	Squaw Creek Watershed
Receiving Waterbody:	Highland Lake
Surface Area:	17.5 acres
Shoreline Length:	0.8 miles
Maximum Depth:	18.5 feet
Average Depth (est.):	9.6 feet
Lake Volume (est.):	163.8 acre-feet
Lake Type:	Glacial Pothole
Watershed Area:	122.5 acres
Major Watershed Land Uses:	Multi-family housing, Wetland
Bottom Ownership:	Village of Hainesville
Management Entities:	Village of Hainesville
Current and Historical Uses:	Historically used for fishing and waterfowl hunting. Currently used for aesthetic enjoyment.
Description of Access:	No public access.

In 2005, Cranberry Lake was chosen to be one of seven “sentinel” lakes in the county that the Lakes Management Unit (LMU) will monitor annually for five years. This report summarizes the water quality sampling results and aquatic plant surveys conducted in 2007 on Cranberry Lake. Similar reports have been written on data collected in 2000, 2005, and 2006 and are available from the LMU. The following report does not cover lake history and discussion of the watershed, as the 2000 and 2005 reports have.

SUMMARY OF WATER QUALITY

Cranberry Lake is listed as an ADID (advanced identification) wetland by the U.S. Environmental Protection Agency. This indicates the lake and the surrounding natural environment have potential for high quality aquatic resources based on water quality and hydrology values.

Water samples were taken once a month at the deepest location in the lake, from May to September (Figure 1). Two samples were taken; one from the upper water layer (epilimnion) at three feet and one from the lower water layer (hypolimnion) three feet above the bottom (Appendix A). They were analyzed for nutrients, solids concentration and other physical parameters (Appendix B; Appendix C). Cranberry Lake was stratified from May until September, with the strongest thermal stratification occurring in July. Thermal stratification is measured in relative thermal resistance to mixing (RTRM). RTRM values that fall below 20 generally allow water within the entire water body to mix freely, while values of 20 and higher generally do not allow layers to mix. The size of this strongly stratified layer usually increases throughout the summer, which was the case in Cranberry Lake.

The average epilimnetic dissolved oxygen (DO) concentration was 5.79 mg/L (Table 1), with the highest reading in May, and the lowest in September. The hypolimnion experienced anoxic conditions (<1.0 mg/L DO) from June through September. DO concentrations in the epilimnion were below 3 mg/L at the surface in September. Because a bathymetric map does not exist for Cranberry Lake, we are unable to calculate the volume of water affected by these low DO concentrations. These low oxygen levels affect fish populations since a healthy bass/sunfish fishery needs concentrations >5.0 mg/L. The creation of a bathymetric map is recommended and the LMU has equipment to do so. The LMU continues to recommend that the lake not be managed as a public fishery, based primarily on the DO conditions.

Suspended solids are made up of any type of solid particles in the water column, including algal cells and sediment. The average epilimnetic total suspended solid (TSS) concentration for Cranberry Lake during the 2007 study was 1.8 mg/L. This is similar to the average concentration in 2005 (1.5 mg/L) and 2006 (1.6 mg/L), and over four times lower than the County median (8.0 mg/L). Cranberry Lake also has the lowest TSS concentration in the Squaw Creek watershed. This is mainly due to the lake being situated at the top of the watershed and having a margin of cattails to filter storm water run off before it enters the lake.

Due to low TSS, Secchi depth (water clarity) in Cranberry Lake was good (Figure 2). The average Secchi depth in 2007 was 9.06 feet, which was a decrease of over one foot since the 2005 sampling season (10.52 feet), and a slight decrease since 2006 (9.33 feet). The slight

decline in water clarity may be due to annual variation or stormwater inputs. The 2007 Secchi depth on Cranberry Lake was still better than the median county Secchi depth (3.28 feet). The deepest Secchi reading was in June (9.84 feet) and the lowest reading was in August (7.87 feet). This poor August reading correlates with the highest TSS concentration of the season (2.3 mg/L). The average conductivity in 2007 was 0.5138 mS/cm, which was a decrease from the 2006 value of 0.6019 mS/cm, and a 35% increase from the 2000 value of 0.3809 mS/cm. The decrease from 2006 to 2007 may have been due to the rain events in 2007 that raised the water level and diluted the ions that make up conductivity. Cranberry Lake was still below the county median (0.8038 mS/cm). Conductivity concentrations varied slightly throughout the Squaw Creek watershed. In 2007, Long Lake had an average value of 0.9066 mS/cm, which is the lake at the bottom of the watershed (Table 2). The road salts used in winter road management runoff enter our water supplies and build up in lakes and most often increase both conductivity and chloride ion (Cl⁻) concentrations, which are correlated (Figure 3). Almost all of the lakes in the county are experiencing increases in conductivity for the same reason. The median Cl⁻ concentration in the county is 158 mg/L, but Cranberry Lake contained less than this concentration in 2007 (78 mg/L). This was a surprising decrease from 2006 (90 mg/L) that correlated to the decrease in conductivity concentrations also seen in 2007. Because Cranberry Lake is a unique natural resource, this decrease in conductivity and Cl⁻ is promising, but again, could have been due to the higher water levels from the August flooding. It is recommended that alternatives to salt be explored.

In 2007, the average epilimnetic total phosphorus (TP) concentration in Cranberry Lake was 0.023 mg/L, which was less than the county median (0.063 mg/L). TP conditions have remained steady since sampling was performed in 2006, 2005 and 2000 when the concentrations were 0.024 mg/L. As a comparison, in 2007, Long Lake had a much higher average TP concentration of 0.103 mg/L. Cranberry Lake had low TP concentrations due in part to it being at the top of the watershed and the cattail fringe that surrounds the entire lake, which helps filter nutrients before they enter the lake. The trophic state of Cranberry Lake in terms of its phosphorus concentration in 2007 was eutrophic ($\geq 50 < 70$), with a TSI_p score of 50, which was also the TSI_p in 2006 and 2005. However, based on Secchi TSI (45), Cranberry Lake is mesotrophic ($\geq 40 < 50$). Therefore, the lake was considered borderline eutrophic. The lake was ranked 19th out of 163 lakes in Lake County based on TP concentrations (Table 3). To help keep the phosphorous level low, the LMU recommends that phosphorous-free fertilizer be used by the area residents.

TSI values along with other water quality parameters can be used to make other analyses based on use impairment indexes established by the Illinois Environmental Protection Agency (IEPA). Most water quality standard impairment assessments were listed as *None*. However, low DO concentrations were the source of impairments. Furthermore, based on IEPA indices, Cranberry Lake has *Partial* support for recreational use and *Full* support for aquatic life use. Based on these indices, this lake is listed as providing *Full* overall use support.

The water level in Cranberry Lake fluctuated in 2007. A stake installed along the east was used to measure the water level each month. Between the May and June sampling the water level increased 2.9 inches. The water level then dropped 1.5 inches from June to July. The water level rose 4.5 inches in August. In September the level decreased 1.9 inches for a seasonal

increase of 4.0 inches. This change in water level was a result of large rains that fell across Lake County during 2007. A total of 23.9 inches of rain was recorded at the Stormwater Management Commission's rain gauge in Round Lake Beach between the May and September sampling.

Figure 1. Water quality sampling point on Cranberry Lake, 2007.

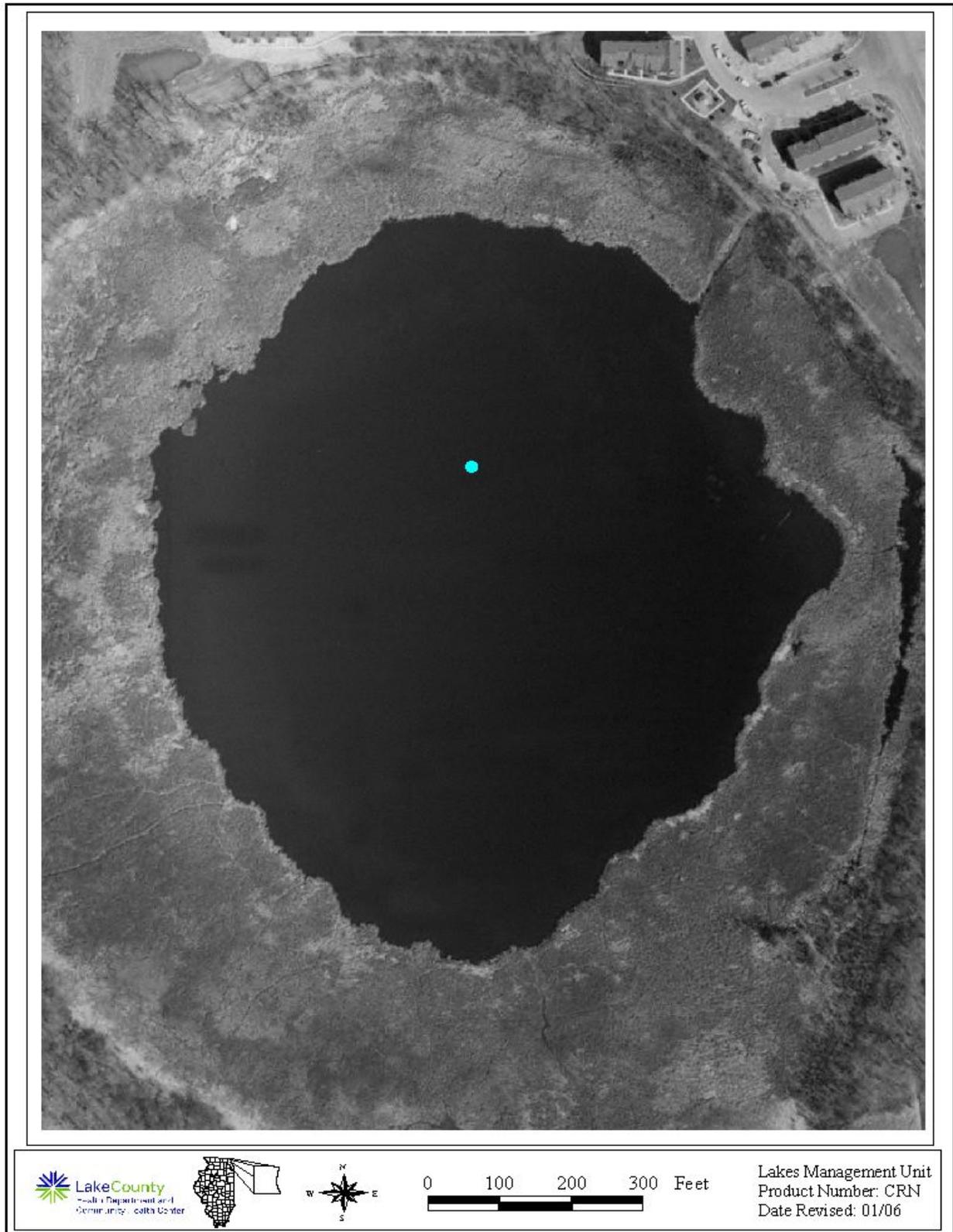


Table 1. Water quality data for Cranberry Lake, 2005, 2006 and 2007.

2007		Epilimnion													
DATE	DEPTH	ALK	TKN	NH ₃ -N	NO ₂ +NO ₃ *	TP	SRP	Cl ⁻	TSS	TS	TVS	SECCHI	COND	pH	DO
09-May	3	139	1.05	<0.1	<0.05	0.025	<0.005	84.0	2.1	352	102	9.19	0.5607	8.01	9.23
13-Jun	3	127	1.04	<0.1	<0.05	0.029	<0.005	82.0	1.3	346	108	9.84	0.5297	8.10	8.04
11-Jul	3	120	1.00	<0.1	<0.05	0.017	<0.005	81.2	1.4	342	116	8.69	0.5148	7.62	5.71
08-Aug	3	111	0.99	<0.1	<0.05	0.025	<0.005	72.8	2.3	306	102	7.87	0.4818	7.56	3.53
12-Sep	3	120	0.98	<0.1	<0.05	0.021	<0.005	71.6	<1.0	315	88	9.71	0.4821	7.18	2.45
Average		123	1.01	<0.1	<0.05	0.023	<0.005	78.3	1.8 ^k	332	103	9.06	0.5138	7.69	5.79

2006		Epilimnion													
DATE	DEPTH	ALK	TKN	NH ₃ -N	NO ₂ +NO ₃ *	TP	SRP	Cl ⁻	TSS	TS	TVS	SECCHI	COND	pH	DO
16-May	3	156	1.44	<0.1	<0.05	0.039	<0.005	88.0	2.4	374	85	7.32	0.6001	7.74	7.47
20-Jun	3	147	1.24	<0.1	<0.05	0.024	<0.005	88.5	1.3	373	100	11.31	0.5994	8.35	7.75
18-Jul	3	142	1.28	<0.1	<0.05	0.018	<0.005	90.5	1.5	385	110	8.86	0.6034	8.05	5.94
15-Aug	3	144	1.36	<0.1	<0.05	0.021	<0.005	94.5	1.3	414	147	9.51	0.6154	7.60	4.31
19-Sep	3	145	1.32	<0.1	<0.05	0.018	<0.005	90.4	1.5	366	91	9.67	0.5914	7.66	4.76
Average		147	1.33	<0.1	<0.05	0.024	<0.005	90.4	1.6	382	107	9.33	0.6019	7.88	6.05

2005		Epilimnion													
DATE	DEPTH	ALK	TKN	NH ₃ -N	NO ₃ -N	TP	SRP	Cl ⁻	TSS	TS	TVS	SECCHI	COND	pH	DO
10-May	3	139	1.31	<0.1	<0.05	0.032	<0.005	72.3	2.2	320	84	8.86	0.5081	7.82	8.06
14-Jun	3	136	1.40	<0.1	<0.05	0.015	<0.005	76.9	1.1	334	99	11.32	0.5362	7.62	5.88
12-Jul	3	147	2.54	<0.1	<0.05	0.019	<0.005	82.9	1.1	399	147	12.14	0.5684	7.44	7.79
09-Aug	3	159	1.43	<0.1	<0.05	0.029	<0.005	86.3	1.5	384	114	9.80	0.6002	8.42	6.64
13-Sep	3	154	1.33	<0.1	<0.05	0.023	<0.005	87.2	<1.0	383	120	10.49	0.5994	8.25	6.65
Average		147	1.60	<0.1	<0.05	0.024	<0.005	81.1	1.5 ^k	364	113	10.52	0.5625	7.91	7.00

Glossary

ALK = Alkalinity, mg/L CaCO ₃	Cl ⁻ = Chloride ions, mg/L
TKN = Total Kjeldahl nitrogen, mg/L	TSS = Total suspended solids, mg/L
NH ₃ -N = Ammonia nitrogen, mg/L	TS = Total solids, mg/L
NO ₃ -N = Nitrate nitrogen, mg/L	TVS = Total volatile solids, mg/L
NO ₂ +NO ₃ = Nitrite and Nitrate nitrogen, mg/L	SECCHI = Secchi disk depth, ft.
TP = Total phosphorus, mg/L	COND = Conductivity, milliSiemens/cm
SRP = Soluble reactive phosphorus, mg/L	DO = Dissolved oxygen, mg/L
TDS = Total dissolved solids, mg/L	

Note: "k" denotes that the actual value is known to be less than the value presented.

NA = Not Applicable

* = Prior to 2006 only Nitrate was analyzed

Table 1. Continued

2007		Hypolimnion													
DATE	DEPTH	ALK	TKN	NH ₃ -N	NO ₂ +NO ₃ *	TP	SRP	Cl ⁻	TSS	TS	TVS	SECCHI	COND	pH	DO
09-May	14	140	1.06	<0.1	<0.05	0.025	<0.005	85.4	1.5	328	71	NA	0.5408	7.26	1.03
13-Jun	16	148	1.64	0.132	<0.05	0.080	<0.005	83.8	11.0	374	112	NA	0.5913	6.93	0.17
11-Jul	15	174	2.43	1.000	<0.05	0.105	0.021	82.2	7.9	386	112	NA	0.6242	6.70	0.23
08-Aug	16	202	4.8	3.910	<0.05	0.254	0.183	84.6	5.1	386	98	NA	0.6784	6.52	0.10
12-Sep	16	194	4.22	3.180	<0.05	0.160	0.112	83.2	3.8	393	99	NA	0.7023	6.42	0.12
Average		172	2.83	2.056 ^k	<0.05	0.125	0.105 ^k	83.8	5.9	373	98	NA	0.6274	6.77	0.33

2006		Hypolimnion													
DATE	DEPTH	ALK	TKN	NH ₃ -N	NO ₂ +NO ₃ *	TP	SRP	Cl ⁻	TSS	TS	TVS	SECCHI	COND	pH	DO
16-May	14	157	1.55	0.152	<0.05	0.043	<0.005	88.3	2.9	375	91	NA	0.6098	7.25	1.36
20-Jun	15	186	3.71	1.840	<0.05	0.115	0.012	88.8	8.2	414	114	NA	0.6843	6.65	0.16
18-Jul	16	240	8.97	7.480	<0.05	0.503	0.458	89.0	6.4	448	120	NA	0.6998	6.56	0.14
15-Aug	14	215	5.81	3.620	<0.05	0.271	0.165	92.0	9.6	450	135	NA	0.7419	6.51	0.16
19-Sep	15	185	5.18	2.950	<0.05	0.246	0.158	92.8	26.0	425	122	NA	0.7564	6.67	0.19
Average		197	5.04	3.208	<0.05	0.236	0.19825 ^k	90.2	10.6	422	116	NA	0.6984	6.73	0.40

2005		Hypolimnion													
DATE	DEPTH	ALK	TKN	NH ₃ -N	NO ₃ -N	TP	SRP	Cl ⁻	TSS	TS	TVS	SECCHI	COND	pH	DO
10-May	14	141	1.21	<0.1	<0.05	0.020	<0.005	72.0	1.7	317	82	NA	0.5040	7.22	6.40
14-Jun	14	146	1.91	<0.1	<0.05	0.045	<0.005	72.6	8.4	335	92	NA	0.5279	6.56	0.12
12-Jul	15	164	1.32	0.205	<0.05	0.153	<0.005	71.2	23.0	377	123	NA	0.5788	6.54	0.18
09-Aug	15	201	3.15	1.500	<0.05	0.152	0.079	74.0	8.6	387	108	NA	0.6023	6.01	0.19
13-Sep	15	198	3.10	1.020	<0.05	0.120	0.031	72.9	10.0	392	108	NA	0.6628	6.50	0.20
Average		170	2.14	0.908 ^k	<0.05	0.098	0.055 ^k	72.5	10.3	362	103	NA	0.5752	6.57	1.42

Glossary

ALK = Alkalinity, mg/L CaCO ₃	Cl ⁻ = Chloride ions, mg/L
TKN = Total Kjeldahl nitrogen, mg/L	TSS = Total suspended solids, mg/L
NH ₃ -N = Ammonia nitrogen, mg/L	TS = Total solids, mg/L
NO ₃ -N = Nitrate nitrogen, mg/L	TVS = Total volatile solids, mg/L
NO ₂ +NO ₃ = Nitrite and Nitrate nitrogen, mg/L	SECCHI = Secchi disk depth, ft.
TP = Total phosphorus, mg/L	COND = Conductivity, milliSiemens/cm
SRP = Soluble reactive phosphorus, mg/L	DO = Dissolved oxygen, mg/L
TDS = Total dissolved solids, mg/L	

Note: "k" denotes that the actual value is known to be less than the value presented.

NA = Not Applicable

* = Prior to 2006 only Nitrate was analyzed

Figure 2. Total suspended solid (TSS) concentrations vs. Secchi depth for Cranberry Lake, 2007.

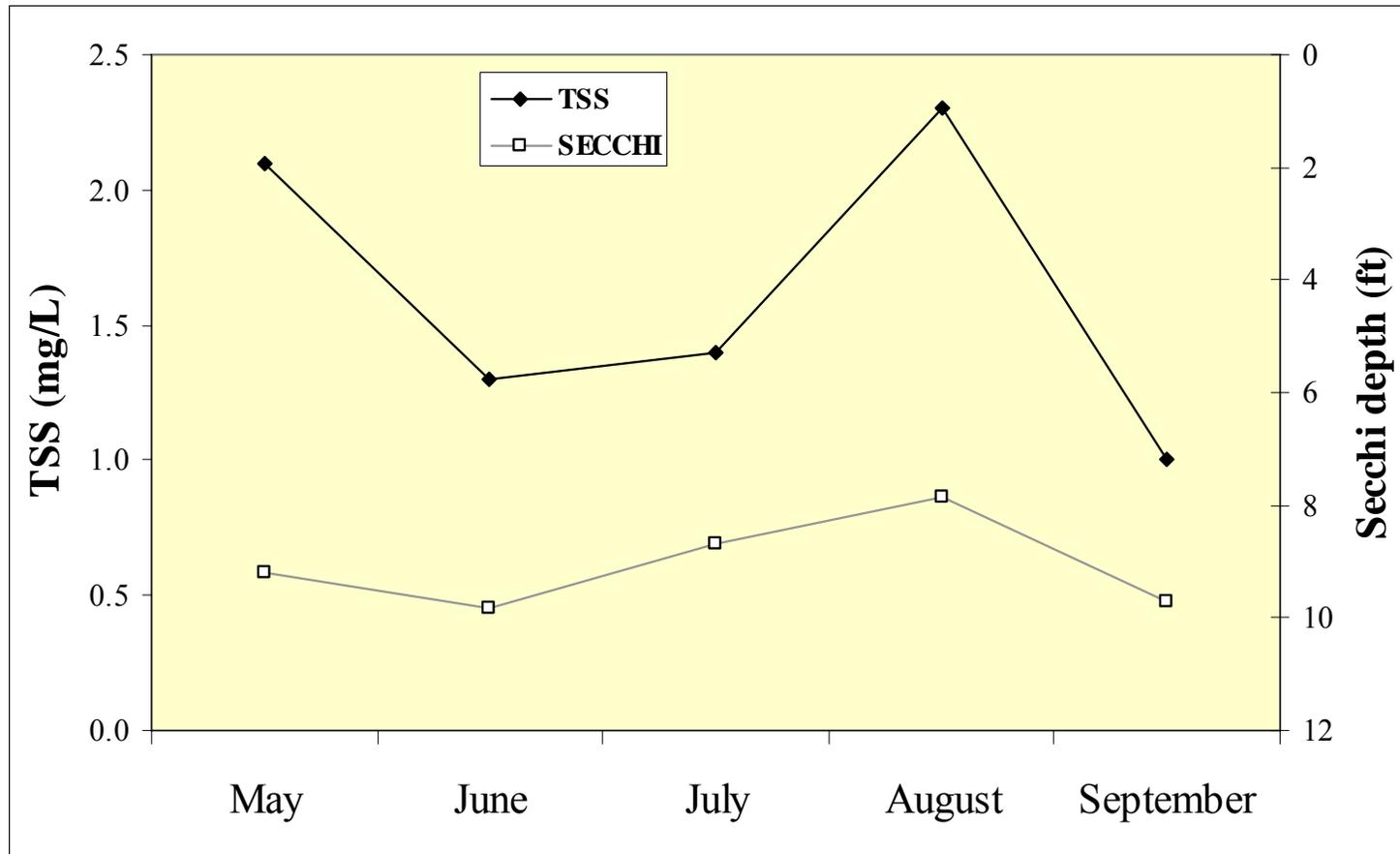


Table 2. Comparison of selected water quality parameters in the Squaw Creek watershed.

	Cranberry Lake	Cranberry Lake	Cranberry Lake	Cranberry Lake	Highland Lake	Highland Lake	Round Lake	Round Lake	Round Lake	Long Lake	Long Lake	Long Lake	Long Lake
Year	2000	2005	2006	2007	1996	2001	1995	1999	2003	2001	2005	2006	2007
Secchi (feet)	10.96	10.52	9.33	9.06	7.98	6.58	7.44	10.32	6.25	4.11	4.18	4.52	3.24
TSS (mg/L)	1.2	1.5	1.6	1.8	2.4	3.3	3.4	2.7	3.5	9.7	10.9	7.2	11.1
TP (mg/L)	0.024	0.024	0.024	0.023	0.023	0.030	0.024	0.015	0.025	0.092	0.076	0.068	0.103
Conductivity (milliSiemens/cm)	0.3809	0.5625	0.6019	0.5138	0.4080	0.5560	0.6290	0.8364	1.073	0.9430	1.0821	1.112	0.9066



Figure 3. Conductivity vs. chloride concentrations in Cranberry Lake, 2007.

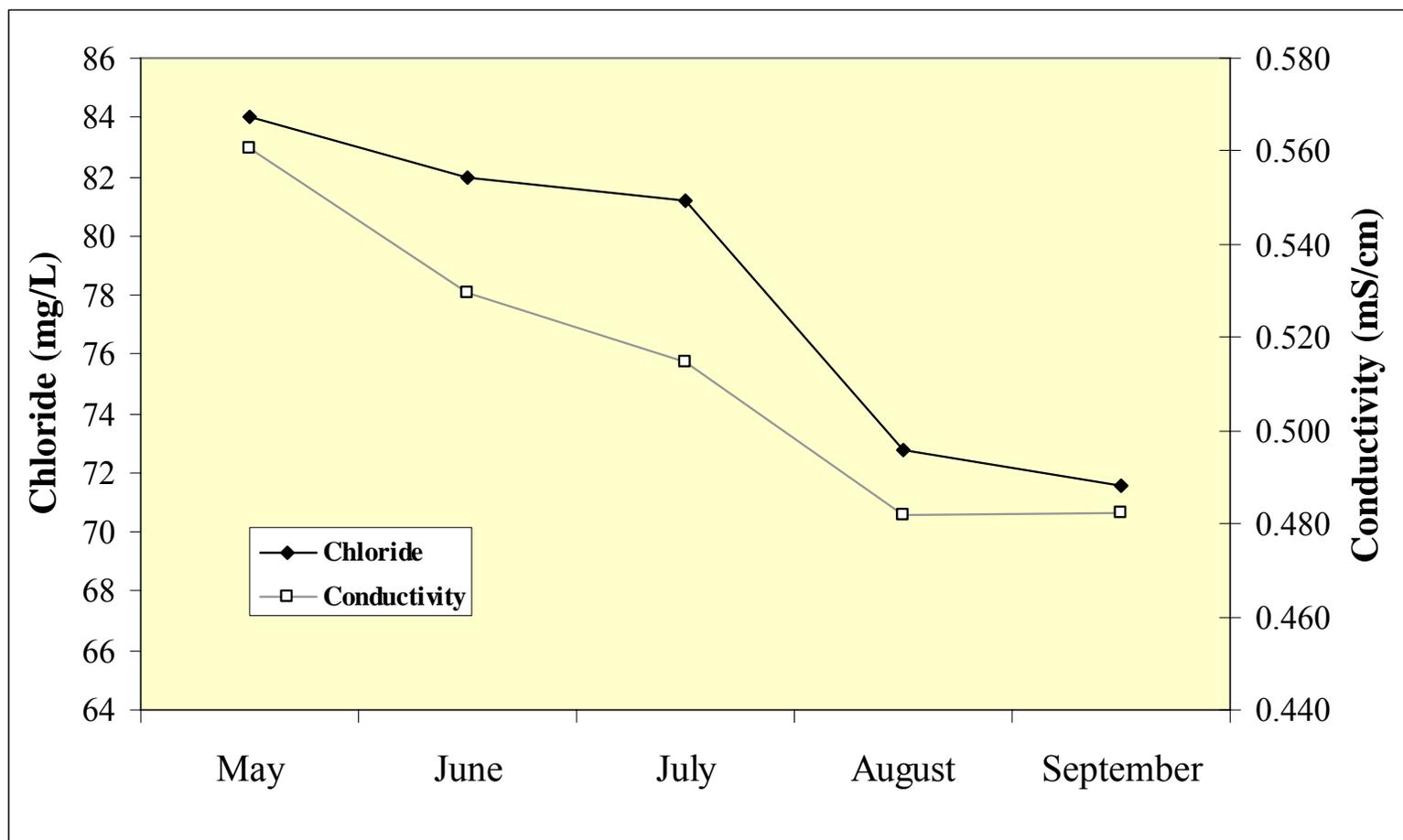


Table 3. Lake County average TSI phosphorous (TSIp) ranking 2000-2007.

RANK	LAKE NAME	TP AVE	TSIp
1	Lake Carina	0.0100	37.35
2	Sterling Lake	0.0100	37.35
3	Independence Grove	0.0114	39.24
4	Sand Pond (IDNR)	0.0132	41.36
5	Cedar Lake	0.0157	41.60
6	Windward Lake	0.0158	43.95
7	Pulaski Pond	0.0180	45.83
8	Timber Lake (North)	0.0180	45.83
9	Fourth Lake	0.0182	45.99
10	West Loon Lake	0.0182	45.99
11	Lake Kathryn	0.0200	47.35
12	Lake of the Hollow	0.0200	47.35
13	Banana Pond	0.0202	47.49
14	Lake Minear	0.0204	47.63
15	Bangs Lake	0.0212	48.17
16	Cross Lake	0.0220	48.72
17	Dog Pond	0.0222	48.85
18	Stone Quarry Lake	0.0230	49.36
19	Cranberry Lake	0.0234	49.61
20	Deep Lake	0.0240	49.98
21	Druce Lake	0.0244	50.22
22	Little Silver Lake	0.0246	50.33
23	Round Lake	0.0254	50.80
24	Lake Leo	0.0256	50.91
25	Dugdale Lake	0.0274	51.89
26	Peterson Pond	0.0274	51.89
27	Lake Miltmore	0.0276	51.99
28	East Loon Lake	0.0280	52.20
29	Lake Zurich	0.0282	52.30
30	Lake Fairfield	0.0296	53.00
31	Gray's Lake	0.0302	53.29
32	Highland Lake	0.0302	53.29
33	Hook Lake	0.0302	53.29
34	Lake Catherine (Site 1)	0.0308	53.57
35	Lambs Farm Lake	0.0312	53.76
36	Old School Lake	0.0312	53.76
37	Sand Lake	0.0316	53.94
38	Sullivan Lake	0.0320	54.13
39	Lake Linden	0.0326	54.39
40	Countryside Lake	0.0332	54.66
41	Gages Lake	0.0338	54.92
42	Hendrick Lake	0.0344	55.17
43	Third Lake	0.0346	55.24
44	Diamond Lake	0.0372	56.30
45	Channel Lake (Site 1)	0.0380	56.60
46	Ames Pit	0.0390	56.98

Table 3. Continued.

RANK	LAKE NAME	TP AVE	TSIp
47	White Lake	0.0408	57.63
48	Sun Lake	0.0410	57.70
49	Potomac Lake	0.0424	58.18
50	Duck Lake	0.0426	58.25
51	Old Oak Lake	0.0428	58.32
52	Deer Lake	0.0434	58.52
53	Schreiber Lake	0.0434	58.52
54	Nielsen Pond	0.0448	58.98
55	Turner Lake	0.0458	59.30
56	Seven Acre Lake	0.0460	59.36
57	Willow Lake	0.0464	59.48
58	Lucky Lake	0.0476	59.85
59	Davis Lake	0.0476	59.85
60	East Meadow Lake	0.0478	59.91
61	College Trail Lake	0.0496	60.45
62	Lake Lakeland Estates	0.0524	61.24
63	Butler Lake	0.0528	61.35
64	West Meadow Lake	0.0530	61.40
65	Heron Pond	0.0545	61.80
66	Little Bear Lake	0.0550	61.94
67	Lucy Lake	0.0552	61.99
68	Lake Christa	0.0576	62.60
69	Lake Charles	0.0580	62.70
70	Crooked Lake	0.0608	63.38
71	Waterford Lake	0.0610	63.43
72	Lake Naomi	0.0616	63.57
73	Lake Tranquility S1	0.0618	63.62
74	Werhane Lake	0.0630	63.89
75	Liberty Lake	0.0632	63.94
76	Countryside Glen Lake	0.0642	64.17
77	Lake Fairview	0.0648	64.30
78	Leisure Lake	0.0648	64.30
79	Tower Lake	0.0662	64.61
80	Wooster Lake	0.0663	64.63
81	St. Mary's Lake	0.0666	64.70
82	Mary Lee Lake	0.0682	65.04
83	Hastings Lake	0.0684	65.08
84	Honey Lake	0.0690	65.21
85	Spring Lake	0.0726	65.94
86	ADID 203	0.0730	66.02
87	Bluff Lake	0.0734	66.10
88	Harvey Lake	0.0766	66.71
89	Broberg Marsh	0.0782	67.01
90	Echo Lake	0.0792	67.19
91	Sylvan Lake	0.0794	67.23
92	Big Bear Lake	0.0806	67.45

Table 3. Continued.

RANK	LAKE NAME	TP AVE	TSIp
93	Petite Lake	0.0834	67.94
94	Timber Lake (South)	0.0848	68.18
95	Lake Marie (Site 1)	0.0850	68.21
96	North Churchill Lake	0.0872	68.58
97	Grand Avenue Marsh	0.0874	68.61
98	Grandwood Park, Site II, Outflow	0.0876	68.65
99	North Tower Lake	0.0878	68.68
100	South Churchill Lake	0.0896	68.97
101	Rivershire Pond 2	0.0900	69.04
102	McGreal Lake	0.0914	69.26
103	International Mine and Chemical Lake	0.0948	69.79
104	Eagle Lake (Site I)	0.0950	69.82
105	Valley Lake	0.0950	69.82
106	Dunns Lake	0.0952	69.85
107	Fish Lake	0.0956	69.91
108	Lochanora Lake	0.0960	69.97
109	Owens Lake	0.0978	70.23
110	Woodland Lake	0.0986	70.35
111	Island Lake	0.0990	70.41
112	McDonald Lake 1	0.0996	70.50
113	Longview Meadow Lake	0.1024	70.90
114	Long Lake	0.1029	70.96
115	Lake Barrington	0.1053	71.31
116	Redwing Slough, Site II, Outflow	0.1072	71.56
117	Lake Forest Pond	0.1074	71.59
118	Bittersweet Golf Course #13	0.1096	71.88
119	Fox Lake (Site 1)	0.1098	71.90
120	Osprey Lake	0.1108	72.04
121	Bresen Lake	0.1126	72.27
122	Round Lake Marsh North	0.1126	72.27
123	Deer Lake Meadow Lake	0.1158	72.67
124	Taylor Lake	0.1184	72.99
125	Columbus Park Lake	0.1226	73.49
126	Nippersink Lake (Site 1)	0.1240	73.66
127	Grass Lake (Site 1)	0.1288	74.21
128	Lake Holloway	0.1322	74.58
129	Lakewood Marsh	0.1330	74.67
130	Summerhill Estates Lake	0.1384	75.24
131	Redhead Lake	0.1412	75.53
132	Forest Lake	0.1422	75.63
133	Antioch Lake	0.1448	75.89
134	Slocum Lake	0.1496	76.36
135	Drummond Lake	0.1510	76.50
136	Pond-a-Rudy	0.1514	76.54
137	Lake Matthews	0.1516	76.56
138	Buffalo Creek Reservoir	0.1550	76.88

Table 3. Continued.

RANK	LAKE NAME	TP AVE	TSIp
139	Pistakee Lake (Site 1)	0.1592	77.26
140	Salem Lake	0.1650	77.78
141	Half Day Pit	0.1690	78.12
142	Lake Eleanor Site II, Outflow	0.1812	79.13
143	Lake Farmington	0.1848	79.41
144	ADID 127	0.1886	79.71
145	Lake Louise Inlet	0.1938	80.10
146	Grassy Lake	0.1952	80.20
147	Dog Bone Lake	0.1990	80.48
148	Redwing Marsh	0.2072	81.06
149	Stockholm Lake	0.2082	81.13
150	Bishop Lake	0.2156	81.63
151	Hidden Lake	0.2236	82.16
152	Fischer Lake	0.2278	82.43
153	Lake Napa Suwe (Outlet)	0.2304	82.59
154	Patski Pond (outlet)	0.2512	83.84
155	Oak Hills Lake	0.2792	85.36
156	Loch Lomond	0.2954	86.18
157	McDonald Lake 2	0.3254	87.57
158	Fairfield Marsh	0.3264	87.61
159	ADID 182	0.3280	87.69
160	Slough Lake	0.4134	91.02
161	Flint Lake Outlet	0.4996	93.75
162	Rasmussen Lake	0.5025	93.84
163	Albert Lake, Site II, outflow	1.1894	106.3

SUMMARY OF AQUATIC MACROPHYTES

An aquatic plant (macrophyte) survey was conducted in May and August of 2007. These sample times allowed the determination of plant growth at the beginning and end of the season. On Cranberry Lake, there were 74 sampling sites in May and 71 sampling sites in August, covering the entire lake (Figure 4; Figure 5). Overall, there were 16 species found in May, with Coontail and White Water Lily having the highest density (found at 59% and 65% of the sites, respectively; Table 4). Flatstem Pondweed, Common Bladderwort and Spatterdock were also abundant and were found at 53%, 51% and 49% of the sites, respectively (Table 5a). This was an increase from 2006 when 15 species were found and 2005 when only 11 species were found. In June, Coontail was the most common species in 2006 and *Chara* spp. was the most common in 2005. A total of 18 species were found in August, Coontail and White Water Lily again having the highest density (found at 63% and 61% of the sites, respectively). Small Bladderwort (found at 59% of the sites) and Common Bladderwort (found at 52% of the sites) were the next most abundant species (Table 5b). This was an increase from 2006 when 14 species were found and 2005 when 16 plant species were found. In August, the Common Bladderwort was the most common species encountered in both 2005 and 2006. In all three years there was a good sized Watershield population on the lake. This is a rare species in Lake County and was at 5% of sample sites in 2005, 12% in 2006 and 9% in 2007. Plants need at least 1% of surface light levels in order to survive. In May 2007, plants were found down to a depth of 17 feet, with a 1% light level depth of 17 feet, which correlates well. Coontail does well at low light levels and was the only species found below 10 feet. In August, the 1% light level depth was around 9 feet, while plants were found at depths of 22 feet. Again, Coontail was the only species found below 10 feet. Plants were found at 97% of the sites in May and at 99% of the sites in August 2007 (Table 5c).

Plant coverage increased from 2005 when 82% and 93% of the sample sites were covered in May and August, respectively. Because deep sites did not have plants in 2005, approximately 65% of the lake had plant coverage. Plant coverage stayed relatively similar from 2006, when 99% of the lake had plants in June and 94% had plants in August. In 2007, 97% of the lake had plants in May and 99% had plants in August. This amount of plants is one reason water clarity was so high in the lake. Nutrients are taken up by macrophytes, leaving a small amount for plankton growth. This results in low algal densities, which could cloud the water. Given the aesthetic use and the high quality natural resource that Cranberry Lake is, no aquatic plant management activities are recommended at this time.

Floristic Quality Index (FQI) is a rapid assessment tool designed to evaluate the closeness of the flora of an area to that of undisturbed conditions. A high FQI number indicates there is a large number of sensitive, high quality plant species present in the lake. Non-native species were also included in the FQI calculations for Lake County lakes. The median FQI for 2000-2007 Lake County lakes is 12.5. Cranberry Lake had a FQI of 30.1 in 2007, which was an increase from 26.6 in 2006. It ranked 9th in 2006 and 3rd in 2007 (Table 6).

Figure 4. Aquatic plant sampling grid illustrating plant density on Cranberry Lake, May 2007.

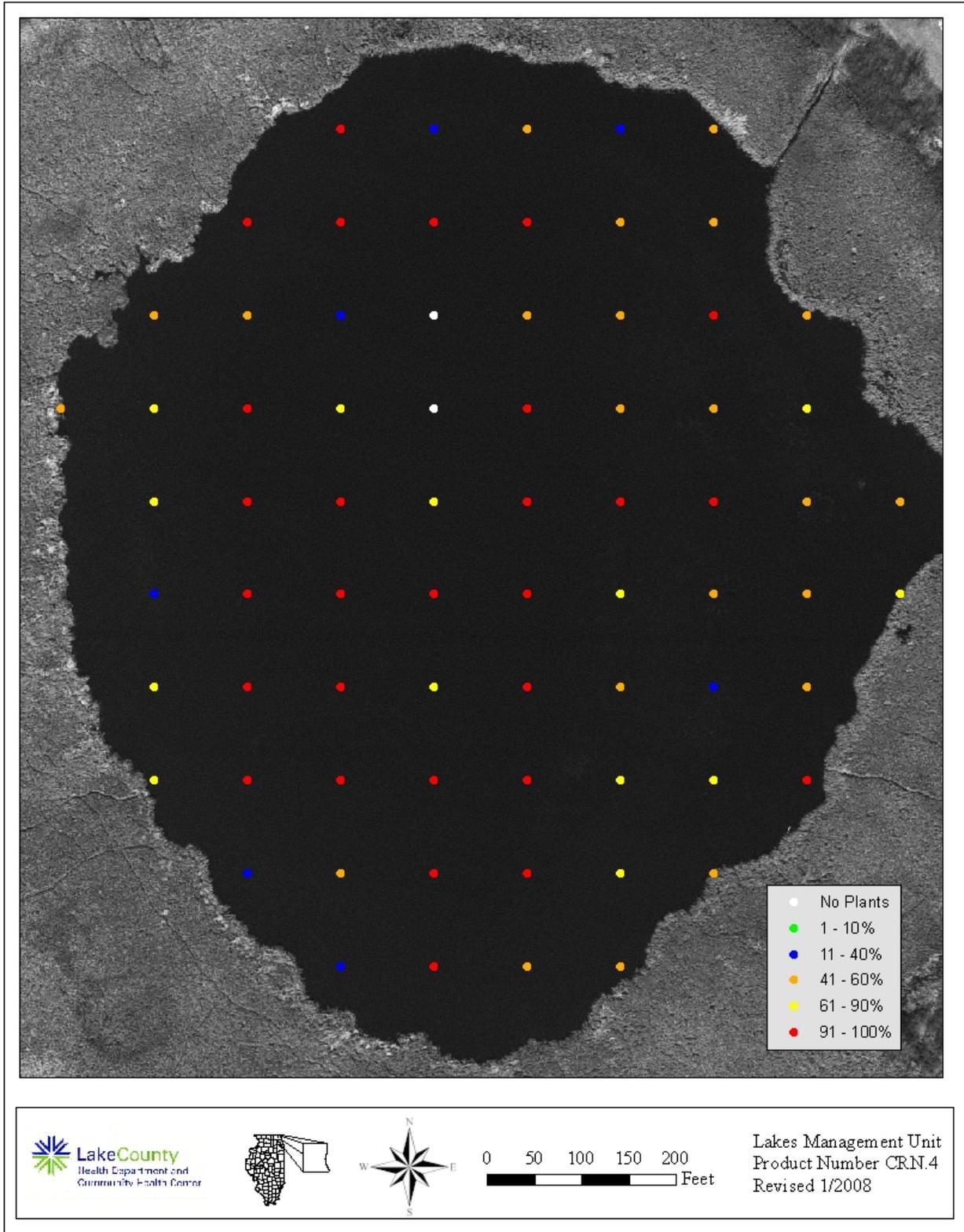


Figure 5. Aquatic plant sampling grid illustrating plant density on Cranberry Lake, August 2007.

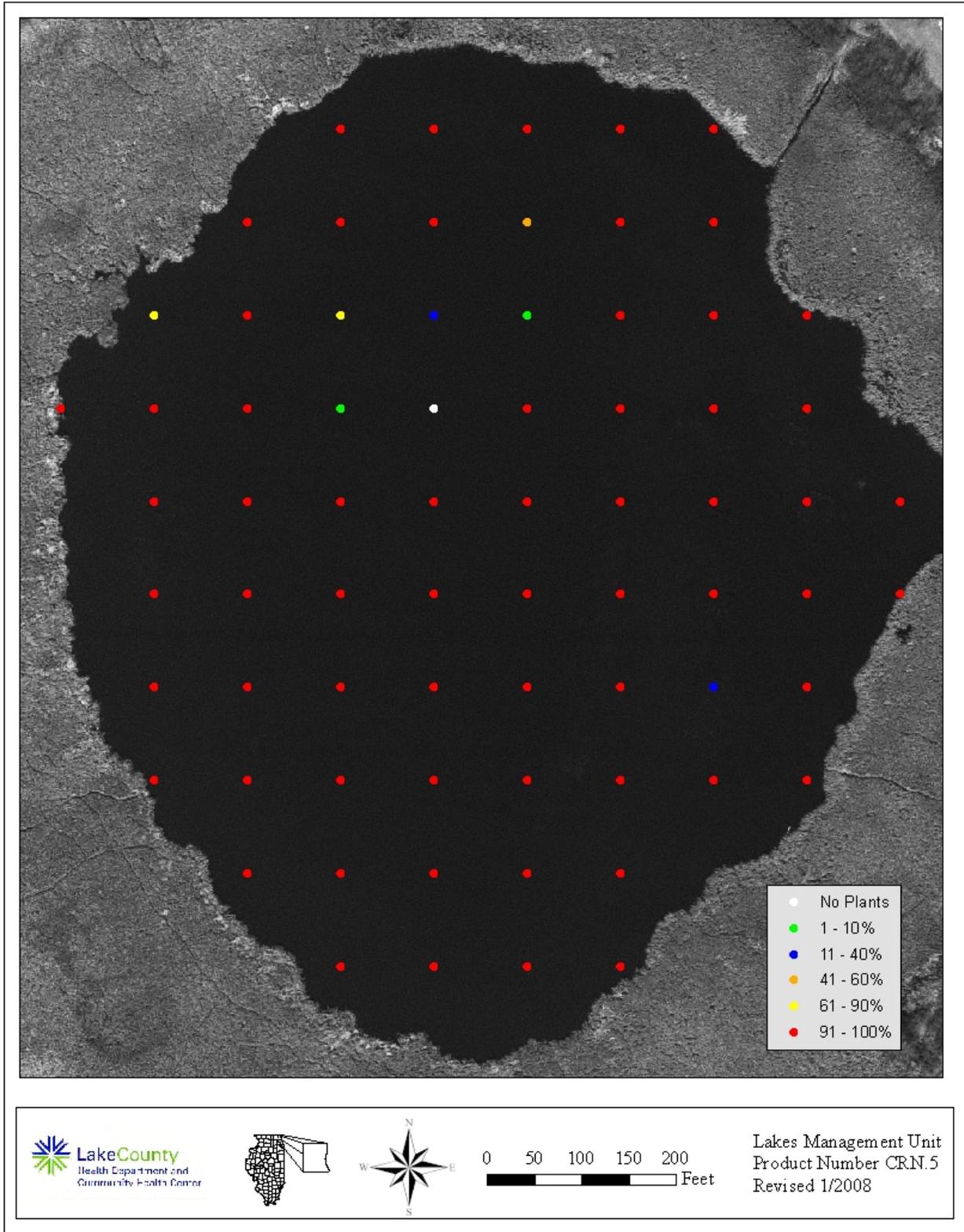


Table 4. Aquatic plant species found in Cranberry Lake, 2007.

Water Shield	<i>Brasenia schreberi</i>
Coontail	<i>Ceratophyllum demersum</i>
Chara (Macro alga)	<i>Chara spp.</i>
Curlyleaf Pondweed	<i>Potamogeton crispus</i>
Illinois Pondweed	<i>Potamogeton illinoensis</i>
Floating-leaf Pondweed	<i>Potamogeton natans</i>
American Pondweed	<i>Potamogeton nodosus</i>
Sago Pondweed	<i>Potamogeton pectinatus</i>
Small Pondweed	<i>Potamogeton pusillus</i>
Flatstem Pondweed	<i>Potamogeton zosteriformis</i>
Northern Watermilfoil	<i>Myriophyllum sibiricum</i>
Slender Naiad	<i>Najas flexilis</i>
Southern Naiad	<i>Najas guadalupensis</i>
Spiny Naiad	<i>Najas marina</i>
Spatterdock	<i>Nuphar variegata</i>
White Water Lily	<i>Nymphaea tuberosa</i>
Softstem Bulrush	<i>Scirpus validus</i>
Humped Bladderwort	<i>Utricularia gibba</i>
Small Bladderwort*	<i>Utricularia minor</i>
Common Bladderwort	<i>Utricularia vulgaris</i>
Vallisneria (Eel grass)	<i>Vallisneria americana</i>

* **Endangered in Illinois**

Table 5a. Aquatic plant species found at the 74 sampling sites on Cranberry Lake in May, 2007. The maximum depth that plants were found was 17.0 feet.

May								
Plant Density	Common Bladderwort	Humped Bladderwort	Small Bladderwort	Chara	Coontail	Curlyleaf Pondweed	Flatstem Pondweed	Floatingleaf Pondweed
Absent	36	69	67	49	30	73	35	73
Present	27	5	7	9	13	1	19	1
Common	7	0	0	5	3	0	10	0
Abundant	4	0	0	5	2	0	9	0
Dominant	0	0	0	6	26	0	1	0
% Plant Occurrence	51	7	9	34	59	1	53	1
Plant Density	Illinois Pondweed	Northern Milfoil	Sago Pondweed	Softstem Bulrush	Spatterdock	Vallisneria	Watershield	White Water Lily
Absent	47	70	60	70	38	68	67	26
Present	19	3	14	4	25	6	4	30
Common	6	0	0	0	9	0	3	17
Abundant	2	1	0	0	2	0	0	1
Dominant	0	0	0	0	0	0	0	0
% Plant Occurrence	36	5	19	5	49	8	9	65

Table 5b. Aquatic plant species found at the 71 sampling sites on Cranberry Lake in August, 2007. The maximum depth that plants were found was 22.0 feet.

August									
Plant Density	American Pondweed	Common Bladderwort	Humped Bladderwort	Small Bladderwort	Chara	Coontail	Flatstem Pondweed	Floatingleaf Pondweed	Illinois Pondweed
Absent	68	34	39	29	60	26	38	69	36
Present	3	31	20	19	5	13	21	2	11
Common	0	5	10	17	4	6	8	0	12
Abundant	0	1	2	6	1	2	3	0	10
Dominant	0	0	0	0	1	24	1	0	2
% Plant Occurrence	4	52	45	59	15	63	46	3	49
Plant Density	Northern Milfoil	Sago Pondweed	Slender Naiad	Southern Naiad	Softstem Bulrush	Spatterdock	Spiny Naiad	Watershield	White Water Lily
Absent	57	56	49	69	67	43	70	65	28
Present	14	12	10	0	4	8	1	4	14
Common	0	3	6	2	0	5	0	1	7
Abundant	0	0	5	0	0	10	0	1	7
Dominant	0	0	1	0	0	5	0	0	15
% Plant Occurrence	20	21	31	3	6	39	1	8	61

Table 5c. Distribution of rake density across all sampled sites on Cranberry Lake.

May		
Rake Density (coverage)	# of Sites	% of Sites
No Plants	2	3%
>0-10%	0	0%
10-40%	7	9%
40-60%	22	30%
60-90%	13	18%
>90%	30	41%
Total Sites with Plants	72	97%
Total # of Sites	74	100%

August		
Rake Density (coverage)	# of Sites	% of Sites
No Plants	1	1%
>0-10%	2	3%
10-40%	2	3%
40-60%	1	1%
60-90%	2	3%
>90%	63	89%
Total Sites with Plants	70	99%
Total # of Sites	71	100%

Table 6. Floristic quality index (FQI) of lakes in Lake County, calculated with exotic species (w/Adventives) and with native species only (native).

RANK	LAKE NAME	FQI (w/A)	FQI (native)
1	Cedar Lake	35.1	37.3
2	Deep Lake	33.9	35.4
3	Cranberry Lake	30.1	31.0
4	Round Lake Marsh North	29.1	29.9
5	East Loon Lake	28.4	29.9
6	Sullivan Lake	28.2	29.7
7	Deer Lake	28.2	29.7
8	Little Silver Lake	27.9	30.0
9	Schreiber Lake	26.8	27.6
10	West Loon Lake	26.0	27.6
11	Cross Lake	25.2	27.8
12	Independence Grove	24.6	27.5
13	Sterling Lake	24.5	26.9
14	Bangs Lake	24.5	26.2
15	Lake Zurich	24.0	26.0
16	Lakewood Marsh	23.8	24.7
17	Lake of the Hollow	23.8	26.2
18	Round Lake	23.5	25.9
19	Fourth Lake	23.0	24.8
20	Druce Lake	22.8	25.2
21	Sun Lake	22.7	24.5
22	Countryside Glen Lake	21.9	22.8
23	Butler Lake	21.4	23.1
24	Duck Lake	21.1	22.9
25	Wooster Lake	20.8	22.6
26	Timber Lake (North)	20.8	22.8
27	Davis Lake	20.5	21.4
28	Broberg Marsh	20.5	21.4
29	ADID 203	20.5	20.5
30	McGreal Lake	20.2	22.1
31	Lake Kathryn	19.6	20.7
32	Redhead Lake	19.3	21.2
33	Owens Lake	19.3	20.2
34	Fish Lake	19.3	21.2
35	Turner Lake	18.6	21.2
36	Salem Lake	18.5	20.2
37	Lake Miltmore	18.4	20.3
38	Hendrick Lake	17.7	17.7
39	Summerhill Estates Lake	17.1	18.0
40	Seven Acre Lake	17.0	15.5
41	Gray's Lake	16.9	19.8
42	Third Lake	16.8	18.7
43	Lake Barrington	16.7	17.7
44	Bresen Lake	16.6	17.8

Table 6. Continued.

Rank	Lake Name	FQI (w/A)	FQI (native)
45	Windward Lake	16.3	17.6
46	Lake Napa Suwe	16.3	17.4
47	Diamond Lake	16.3	17.4
48	Long Lake	16.1	18.0
49	Dog Bone Lake	15.7	15.7
50	Redwing Slough	15.6	16.6
51	Osprey Lake	15.5	17.3
52	Lake Fairview	15.2	16.3
53	Heron Pond	15.1	15.1
54	North Churchill Lake	15.0	15.0
55	Lake Tranquility (S1)	15.0	17.0
56	Island Lake	14.7	16.6
57	Dog Training Pond	14.7	15.9
58	Highland Lake	14.5	16.7
59	Taylor Lake	14.3	16.3
60	Grand Avenue Marsh	14.3	16.3
61	Eagle Lake (S1)	14.0	15.1
62	Dugdale Lake	14.0	15.1
63	Longview Meadow Lake	13.9	13.9
64	Hook Lake	13.4	15.5
65	Bishop Lake	13.4	15.0
66	Ames Pit	13.4	15.5
67	Old School Lake	13.1	15.1
68	McDonald Lake 2	13.1	14.3
69	Mary Lee Lake	13.1	15.1
70	Buffalo Creek Reservoir	13.1	14.3
71	White Lake	12.7	14.7
72	Timber Lake (South)	12.7	14.7
73	Old Oak Lake	12.7	14.7
74	Dunn's Lake	12.7	13.9
75	Stone Quarry Lake	12.5	12.5
76	Sand Lake	12.5	14.8
77	Hastings Lake	12.5	14.8
78	Echo Lake	12.5	14.8
79	Stockholm Lake	12.1	13.5
80	Pond-A-Rudy	12.1	12.1
81	Lambs Farm Lake	12.1	14.3
82	Lake Leo	12.1	14.3
83	Lake Carina	12.1	14.3
84	Honey Lake	12.1	14.3
85	Lake Matthews	12.0	12.0
86	Harvey Lake	11.8	13.0
87	Flint Lake	11.8	13.0
88	Rivershire Pond 2	11.5	13.3
89	Lake Linden	11.3	11.3
90	Lake Charles	11.3	13.4

Table 6. Continued.

Rank	Lake Name	FQI (w/A)	FQI (native)
91	Antioch Lake	11.3	13.4
92	Pulaski Pond	11.2	12.5
93	Lake Naomi	11.2	12.5
94	West Meadow Lake	11.0	11.0
95	Tower Lake	11.0	11.0
96	Redwing Marsh	11.0	11.0
97	Lake Minear	11.0	13.9
98	Nielsen Pond	10.7	12.0
99	Lake Holloway	10.6	10.6
100	Countryside Lake	10.5	12.1
101	Crooked Lake	10.2	12.5
102	Lake Lakeland Estates	10.0	11.5
103	College Trail Lake	10.0	10.0
104	Valley Lake	9.9	9.9
105	Werhane Lake	9.8	12.0
106	Little Bear Lake	9.5	11.0
107	Big Bear Lake	9.5	11.0
108	Loch Lomond	9.4	12.1
109	Sylvan Lake	9.2	9.2
110	Columbus Park Lake	9.2	9.2
111	Lake Fairfield	9.0	10.4
112	Grandwood Park Lake	9.0	11.0
113	Fischer Lake	9.0	11.0
114	McDonald Lake 1	8.9	10.0
115	South Churchill Lake	8.5	8.5
116	Lucy Lake	8.5	9.8
117	Lake Farmington	8.5	9.8
118	Lake Christa	8.5	9.8
119	East Meadow Lake	8.5	8.5
120	Woodland Lake	8.1	9.9
121	Bittersweet Golf Course #13	8.1	8.1
122	Lake Louise	7.5	8.7
123	Lake Eleanor	7.5	8.7
124	Fairfield Marsh	7.5	8.7
125	Banana Pond	7.5	9.2
126	Albert Lake	7.5	8.7
127	Slough Lake	7.1	7.1
128	Rasmussen Lake	7.1	7.1
129	Patski Pond	7.1	7.1
130	Lucky Lake	7.0	7.0
131	Lake Forest Pond	6.9	8.5
132	Leisure Lake	6.4	9.0
133	Peterson Pond	6.0	8.5
134	Slocum Lake	5.8	7.1
135	Grassy Lake	5.8	7.1
136	Gages Lake	5.8	10.0

Table 6. Continued.

Rank	Lake Name	FQI (w/A)	FQI (native)
137	Deer Lake Meadow Lake	5.2	6.4
138	Oak Hills Lake	5.0	5.0
139	Liberty Lake	5.0	5.0
140	IMC Lake	5.0	7.1
141	Drummond Lake	5.0	7.1
142	ADID 127	5.0	5.0
143	Sand Pond (IDNR)	3.5	5.0
144	Forest Lake	3.5	5.0
145	Half Day Pit	2.9	5.0
146	Lochanora Lake	2.5	5.0
147	Willow Lake	0.0	0.0
148	Waterford Lake	0.0	0.0
149	St. Mary's Lake	0.0	0.0
150	Potomac Lake	0.0	0.0
151	North Tower Lake	0.0	0.0
152	Hidden Lake	0.0	0.0
	<i>Mean</i>	13.6	14.9
	<i>Median</i>	12.5	14.3

**APPENDIX A. METHODS FOR FIELD DATA COLLECTION AND
LABORATORY ANALYSES**

Water Sampling and Laboratory Analyses

Two water samples were collected once a month from May through September. Sample locations were at the deepest point in the lake (see sample site map), three feet below the surface, and 3 feet above the bottom. Samples were collected with a horizontal Van Dorn water sampler. Approximately three liters of water were collected for each sample for all lab analyses. After collection, all samples were placed in a cooler with ice until delivered to the Lake County Health Department lab, where they were refrigerated. Analytical methods for the parameters are listed in Table A1. Except nitrate nitrogen, all methods are from the Eighteenth Edition of Standard Methods, (eds. American Public Health Association, American Water Works Association, and Water Pollution Control Federation, 1992). Methodology for nitrate nitrogen was taken from the 14th edition of Standard Methods. Dissolved oxygen, temperature, conductivity and pH were measured at the deep hole with a Hydrolab DataSonde® 4a. Photosynthetic Active Radiation (PAR) was recorded using a LI-COR® 192 Spherical Sensor attached to the Hydrolab DataSonde® 4a. Readings were taken at the surface and then every two feet until reaching the bottom.

Plant Sampling

In order to randomly sample each lake, mapping software (ArcMap 9.1) overlaid a grid pattern onto a 2006 aerial photo of Lake County and placed points 60 or 30 meters apart, depending on lake size. Plants were sampled using a garden rake fitted with hardware cloth. The hardware cloth surrounded the rake tines and is tapered two feet up the handle. A rope was tied to the end of the handle for retrieval. At designated sampling sites, the rake was tossed into the water, and using the attached rope, was dragged across the bottom, toward the boat. After pulling the rake into the boat, plant coverage was assessed for overall abundance. Then plants were individually identified and placed in categories based on coverage. Plants that were not found on the rake but were seen in the immediate vicinity of the boat at the time of sampling were also recorded. Plants difficult to identify in the field were placed in plastic bags and identified with plant keys after returning to the office. The depth of each sampling location was measured either by a hand-held depth meter, or by pushing the rake straight down and measuring the depth along the rope or rake handle. One-foot increments were marked along the rope and rake handle to aid in depth estimation.

Wildlife Assessment

Species of wildlife were noted during visits to each lake. When possible, wildlife was identified to species by sight or sound. However, due to time constraints, collection of quantitative information was not possible. Thus, all data should be considered anecdotal. Some of the species on the list may have only been seen once, or were spotted during their migration through the area.

Table A1. Analytical methods used for water quality parameters.

<i>Parameter</i>	<i>Method</i>
Temperature	Hydrolab DataSonde® 4a or YSI 6600 Sonde®
Dissolved oxygen	Hydrolab DataSonde® 4a or YSI 6600 Sonde®
Nitrate and Nitrite nitrogen	USEPA 353.2 rev. 2.0 EPA-600/R-93/100 Detection Limit = 0.05 mg/L
Ammonia nitrogen	SM 18 th ed. Electrode method, #4500 NH ₃ -F Detection Limit = 0.1 mg/L
Total Kjeldahl nitrogen	SM 18 th ed, 4500-N _{org} C Semi-Micro Kjeldahl, plus 4500 NH ₃ -F Detection Limit = 0.5 mg/L
pH	Hydrolab DataSonde® 4a, or YSI 6600 Sonde® Electrometric method
Total solids	SM 18 th ed, Method #2540B
Total suspended solids	SM 18 th ed, Method #2540D Detection Limit = 0.5 mg/L
Chloride	SM 18 th ed, Method #4500C1-D
Total volatile solids	SM 18 th ed, Method #2540E, from total solids
Alkalinity	SM 18 th ed, Method #2320B, potentiometric titration curve method
Conductivity	Hydrolab DataSonde® 4a or YSI 6600 Sonde®
Total phosphorus	SM 18 th ed, Methods #4500-P B 5 and #4500-P E Detection Limit = 0.01 mg/L
Soluble reactive phosphorus	SM 18 th ed, Methods #4500-P B 1 and #4500-P E Detection Limit = 0.005 mg/L
Clarity	Secchi disk
Color	Illinois EPA Volunteer Lake Monitoring Color Chart
Photosynthetic Active Radiation (PAR)	Hydrolab DataSonde® 4a or YSI 6600 Sonde®, LI-COR® 192 Spherical Sensor

**APPENDIX B. MULTI-PARAMETER DATA FOR
CRANBERRY LAKE IN 2007.**

Cranberry Lake 2007 Multiparameter data

Date MMDDYY	Time HHMMSS	Text								Depth of Light Meter feet	% Light	
		Depth feet	Dep25 feet	Temp øC	DO mg/l	DO% Sat	SpCond mS/cm	pH Units	PAR æE/s/mý		Transmission Average	Extinction Coefficient
50907	71217	0	0.52	19.85	8.34	94.3	0.5569	7.92	824	Surface		0.587
50907	71317	1	1.03	19.86	8.23	93.0	0.5573	7.93	776	Surface	100%	
50907	71429	2	1.99	19.85	8.25	93.2	0.5570	7.94	257	0.24	31%	4.604
50907	71533	3	2.99	18.85	9.23	102.2	0.5607	8.01	184	1.24	22%	0.269
50907	71639	4	4.00	17.58	8.78	94.7	0.5545	7.92	148	2.25	18%	0.097
50907	71741	6	6.02	16.03	8.73	91.2	0.5530	7.90	87	4.27	11%	0.124
50907	71833	8	8.05	14.42	7.68	77.5	0.5545	7.75	63	6.30	8%	0.051
50907	71938	10	10.04	13.25	6.26	61.6	0.5486	7.62	49	8.29	6%	0.030
50907	72107	12	12.03	10.75	2.66	24.7	0.5367	7.39	32	10.28	4%	0.041
50907	72227	14	14.01	9.12	1.03	9.3	0.5408	7.26	29	12.26	4%	0.008
50907	72341	16	15.97	7.34	0.28	2.4	0.5513	7.14	13	14.22	2%	0.056

Date	Time	Text	Depth	Dep25	Temp	DO	DO%	SpCond	pH	PAR	Depth of Light	% Light	Extinction
MMDDYY	HHMMSS		feet	feet	øC	mg/l	Sat	mS/cm	Units	æE/s/mý	Meter	Average	Coefficient
61307	72541		0	0.50	23.33	7.78	93.4	0.5305	8.05	3394	Surface		0.494
61307	72628		1	1.02	23.34	7.88	94.6	0.5296	8.06	3245	Surface	100%	
61307	72719		2	2.08	23.33	7.84	94.1	0.5300	8.05	1007	0.33	31%	3.546
61307	72817		3	3.06	23.30	8.04	96.5	0.5297	8.10	784	1.31	24%	0.191
61307	72925		4	4.05	23.25	8.22	98.5	0.5294	8.17	764	2.30	24%	0.011
61307	73119		6	6.08	21.71	3.44	40.1	0.5452	7.51	272	4.33	8%	0.239
61307	73216		8	8.11	19.74	2.00	22.4	0.5653	7.43	134	6.36	4%	0.111
61307	73334		10	10.06	17.05	0.65	6.8	0.5659	7.32	83	8.31	3%	0.058
61307	73501		12	12.02	13.81	0.26	2.6	0.5567	7.23	49	10.27	2%	0.051
61307	73604		14	14.02	10.78	0.20	1.8	0.5690	7.07	18	12.27	0.6%	0.082
61307	73712		16	16.04	8.95	0.17	1.5	0.5913	6.93	2	14.29	0.1%	0.154
61307	73819		18	17.98	8.11	0.12	1.1	0.7587	6.51	0	16.23		

Cranberry		Text								Depth of Light	% Light	Extinction Coefficient 0.833
Date MMDDYY	Time HHMMSS	Depth feet	Dep25 feet	Temp øC	DO mg/l	DO% Sat	SpCond mS/cm	pH Units	PAR æE/s/mý	Meter feet	Transmission Average	
71107	72457	0	0.46	23.85	5.80	67.7	0.5143	7.72	3352	Surface		
71107	72552	1	1.07	23.84	5.67	66.2	0.5148	7.68	3199	Surface	100%	
71107	72706	2	2.02	23.88	5.57	65.0	0.5166	7.65	905	0.27	28%	4.677
71107	72815	3	3.07	23.91	5.71	66.7	0.5148	7.62	353	1.32	11%	0.713
71107	72912	4	4.01	23.91	5.65	66.0	0.5155	7.60	377	2.26	12%	-0.029
71107	73049	6	6.04	23.90	5.51	64.4	0.5168	7.59	188	4.29	6%	0.162
71107	73219	8	8.07	21.99	0.48	5.4	0.5460	7.23	116	6.32	4%	0.076
71107	73352	10	10.07	19.11	0.29	3.1	0.5639	7.17	46	8.32	1.4%	0.111
71107	73534	12	12.04	15.47	0.26	2.5	0.5810	6.99	13	10.29	0.4%	0.123
71107	73701	14	14.06	11.86	0.25	2.3	0.5951	6.79	1	12.31		
71107	73843	16	16.08	9.64	0.22	1.9	0.6533	6.60	0	14.33		
71107	74006	18	18.06	8.88	0.17	1.5	0.7092	6.44	0	16.31		

Date	Time	Text	Depth	Dep25	Temp	DO	DO%	SpCond	pH	PAR	Depth of Light	% Light	Extinction
MMDDYY	HHMMSS		feet	feet	øC	mg/l	Sat	mS/cm	Units	æE/s/mý	Meter	Average	Coefficient
80807	73807		0	0.39	26.40	5.12	62.7	0.4719	8.02	3147	Surface		0.733
80807	73856		1	1.05	26.41	5.02	61.6	0.4716	7.88	3107	Surface	100%	
80807	74020		2	2.11	26.10	4.52	55.2	0.4750	7.71	736	0.36	24%	4.001
80807	74120		3	3.05	25.26	3.53	42.4	0.4818	7.56	459	1.30	15%	0.363
80807	74235		4	4.02	24.68	1.95	23.1	0.5035	7.43	316	2.27	10%	0.164
80807	74406		6	6.07	23.61	0.30	3.5	0.4989	7.28	130	4.32	4%	0.206
80807	74535		8	8.05	22.54	0.18	2.1	0.5275	7.15	38	6.30	1.2%	0.195
80807	74709		10	10.12	20.17	0.14	1.5	0.5729	6.99	12	8.37	0.4%	0.138
80807	74834		12	12.05	17.01	0.12	1.2	0.6001	6.79	6	10.30	0.2%	0.067
80807	74934		14	14.03	13.44	0.12	1.2	0.6116	6.66	0	12.28		
80807	75104		16	16.00	10.91	0.10	0.9	0.6784	6.52	0	14.25		
80807	75244		18	18.02	9.44	0.11	0.9	0.7724	6.33	0	16.27		

Date	Time	Text	Depth	Dep25	Temp	DO	DO%	SpCond	pH	PAR	Depth of Light	% Light	Extinction
MMDDYY	HHMMSS		feet	feet	øC	mg/l	Sat	mS/cm	Units	æE/s/mý	Meter	Transmission	Coefficient
											feet	Average	0.666
91207	71619		0	0.45	18.66	2.74	28.7	0.4812	7.28	3025	Surface		
91207	71730		1	1.00	18.71	2.57	27.0	0.4823	7.24	2701	Surface	100%	
91207	71840		2	2.01	18.71	2.46	25.9	0.4821	7.19	818	0.26	30%	4.594
91207	71925		3	2.99	18.72	2.45	25.7	0.4821	7.18	106	1.24	4%	1.648
91207	72012		4	4.01	18.72	2.42	25.4	0.4818	7.16	44	2.26	2%	0.389
91207	72055		5	5.02	18.71	2.42	25.4	0.4814	7.15	146	3.27	5%	-0.367
91207	72138		6	6.02	18.71	2.44	25.6	0.4814	7.14	81	4.27	3%	0.138
91207	72235		7	6.98	18.70	2.43	25.5	0.4815	7.13	67	5.23	2%	0.036
91207	72325		8	7.98	18.70	2.46	25.8	0.4805	7.13	41	6.23	2%	0.079
91207	72422		9	8.98	18.57	2.83	29.6	0.4779	7.16	22	7.23	0.8%	0.086
91207	72507		10	10.00	18.29	1.59	16.5	0.5042	6.98	19	8.25	0.7%	0.018
91207	72615		11	11.00	17.87	0.26	2.7	0.5891	6.70	13	9.25	0.5%	0.041
91207	72741		12	12.01	16.66	0.17	1.7	0.6087	6.58	7	10.26	0.3%	0.060
91207	72816		13	12.99	15.72	0.15	1.5	0.6206	6.53	4	11.24		
91207	72856		14	14.04	14.47	0.14	1.4	0.6547	6.48	1	12.29		
91207	72937		15	14.95	13.43	0.14	1.3	0.6819	6.45	1	13.20		
91207	73024		16	16.01	12.21	0.12	1.1	0.7023	6.42	0	14.26		
91207	73103		17	16.99	11.31	0.13	1.1	0.7350	6.38	0	15.24		
91207	73145		18	18.00	10.87	0.12	1.1	0.7645	6.33	1	16.25		

**APPENDIX C. INTERPRETING YOUR LAKE'S WATER QUALITY
DATA.**

Lakes possess a unique set of physical and chemical characteristics that will change over time. These in-lake water quality characteristics, or parameters, are used to describe and measure the quality of lakes, and they relate to one another in very distinct ways. As a result, it is virtually impossible to change any one component in or around a lake without affecting several other components, and it is important to understand how these components are linked.

The following pages will discuss the different water quality parameters measured by Lake County Health Department staff, how these parameters relate to each other, and why the measurement of each parameter is important. The median values (the middle number of the data set, where half of the numbers have greater values, and half have lesser values) of data collected from Lake County lakes from 2000-2005 will be used in the following discussion.

Temperature and Dissolved Oxygen:

Water temperature fluctuations will occur in response to changes in air temperatures, and can have dramatic impacts on several parameters in the lake. In the spring and fall, lakes tend to have uniform, well-mixed conditions throughout the water column (surface to the lake bottom). However, during the summer, deeper lakes will separate into distinct water layers. As surface water temperatures increase with increasing air temperatures, a large density difference will form between the heated surface water and colder bottom water. Once this difference is large enough, these two water layers will separate and generally will not mix again until the fall. At this time the lake is thermally stratified. The warm upper water layer is called the *epilimnion*, while the cold bottom water layer is called the *hypolimnion*. In some shallow lakes, stratification and destratification can occur several times during the summer. If this occurs the lake is described as polymictic. Thermal stratification also occurs to a lesser extent during the winter, when warmer bottom water becomes separated from ice-forming water at the surface until mixing occurs during spring ice-out.

Monthly temperature profiles were established on each lake by measuring water temperature every foot (lakes ≤ 15 feet deep) or every two feet (lakes > 15 feet deep) from the lake surface to the lake bottom. These profiles are important in understanding the distribution of chemical/biological characteristics and because increasing water temperature and the establishment of thermal stratification have a direct impact on dissolved oxygen (DO) concentrations in the water column. If a lake is shallow and easily mixed by wind, the DO concentration is usually consistent throughout the water column. However, shallow lakes are typically dominated by either plants or algae, and increasing water temperatures during the summer speeds up the rates of photosynthesis and decomposition in surface waters. When many of the plants or algae die at the end of the growing season, their decomposition results in heavy oxygen consumption and can lead to an oxygen crash. In deeper, thermally stratified lakes, oxygen production is greatest in the top portion of the lake, where sunlight drives photosynthesis, and oxygen consumption is greatest near the bottom of a lake, where sunken organic matter accumulates and decomposes. The oxygen difference between the top and bottom water layers can be dramatic, with plenty of oxygen near the surface, but practically none near the bottom. The oxygen profiles measured during the water quality study can illustrate if

this is occurring. This is important because the absence of oxygen (anoxia) near the lake bottom can have adverse effects in eutrophic lakes resulting in the chemical release of phosphorus from lake sediment and the production of hydrogen sulfide (rotten egg smell) and other gases in the bottom waters. Low oxygen conditions in the upper water of a lake can also be problematic since all aquatic organisms need oxygen to live. Some oxygen may be present in the water, but at too low a concentration to sustain aquatic life. Oxygen is needed by all plants, virtually all algae and for many chemical reactions that are important in lake functioning. Most adult sport-fish such as largemouth bass and bluegill require at least 3 mg/L of DO in the water to survive. However, their offspring require at least 5 mg/L DO as they are more sensitive to DO stress. When DO concentrations drop below 3 mg/L, rough fish such as carp and green sunfish are favored and over time will become the dominant fish species.

External pollution in the form of oxygen-demanding organic matter (i.e., sewage, lawn clippings, soil from shoreline erosion, and agricultural runoff) or nutrients that stimulate the growth of excessive organic matter (i.e., algae and plants) can reduce average DO concentrations in the lake by increasing oxygen consumption. This can have a detrimental impact on the fish community, which may be squeezed into a very small volume of water as a result of high temperatures in the epilimnion and low DO levels in the hypolimnion.

Nutrients:

Phosphorus:

For most Lake County lakes, phosphorus is the nutrient that limits plant and algae growth. This means that any addition of phosphorus to a lake will typically result in algae blooms or high plant densities during the summer. The source of phosphorus to a lake can be external or internal (or both). External sources of phosphorus enter a lake through point (i.e., storm pipes and wastewater discharge) and non-point runoff (i.e., overland water flow). This runoff can pick up large amounts of phosphorus from agricultural fields, septic systems or impervious surfaces before it empties into the lake.

Internal sources of phosphorus originate within the lake and are typically linked to the lake sediment. In lakes with high oxygen levels (oxic), phosphorus can be released from the sediment through plants or sediment resuspension. Plants take up sediment-bound phosphorus through their roots, releasing it in small amounts to the water column throughout their life cycles, and in large amounts once they die and begin to decompose. Sediment resuspension can occur through biological or mechanical means. Bottom-feeding fish, such as common carp and black bullhead can release phosphorus by stirring up bottom sediment during feeding activities and can add phosphorus to a lake through their fecal matter. Sediment resuspension, and subsequent phosphorus release, can also occur via wind/wave action or through the use of artificial aerators, especially in shallow lakes. In lakes that thermally stratify, internal phosphorus release can occur from the sediment through chemical means. Once oxygen is depleted (anoxia) in the hypolimnion, chemical reactions occur in which phosphorus bound to iron complexes in the sediment becomes soluble and is released into the water column. This phosphorus is trapped in the hypolimnion and is unavailable to algae until fall turnover, and can cause algae blooms once

it moves into the sunlit surface water at that time. Accordingly, many of the lakes in Lake County are plagued by dense algae blooms and excessive, exotic plant coverage, which negatively affect DO levels, fish communities and water clarity.

Lakes with an average phosphorus concentration greater than 0.05 mg/L are considered nutrient rich. The median near surface total phosphorus (TP) concentration in Lake County lakes from 2000-2005 is 0.063 mg/L and ranged from a non-detectable minimum of <0.010 mg/L on five lakes to a maximum of 3.880 mg/L on Albert Lake. The median anoxic TP concentration in Lake County lakes from 2000-2005 was 0.174 mg/L and ranged from a minimum of 0.012 mg/L in West Loon Lake to a maximum of 3.880 mg/L in Taylor Lake.

The analysis of phosphorus also included soluble reactive phosphorus (SRP), a dissolved form of phosphorus that is readily available for plant and algae growth. SRP is not discussed in great detail in most of the water quality reports because SRP concentrations vary throughout the season depending on how plants and algae absorb and release it. It gives an indication of how much phosphorus is available for uptake, but, because it does not take all forms of phosphorus into account, it does not indicate how much phosphorus is truly present in the water column. TP is considered a better indicator of a lake's nutrient status because its concentrations remain more stable than soluble reactive phosphorus. However, elevated SRP levels are a strong indicator of nutrient problems in a lake.

Nitrogen:

Nitrogen is also an important nutrient for plant and algae growth. Sources of nitrogen to a lake vary widely, ranging from fertilizer and animal wastes, to human waste from sewage treatment plants or failing septic systems, to groundwater, air and rainfall. As a result, it is very difficult to control or reduce nitrogen inputs to a lake. Different forms of nitrogen are present in a lake under different oxic conditions. NH_4^+ (ammonium) is released from decomposing organic material under anoxic conditions and accumulates in the hypolimnion of thermally stratified lakes. If NH_4^+ comes into contact with oxygen, it is immediately converted to NO_2^- (nitrite) which is then oxidized to NO_3^- (nitrate). Therefore, in a thermally stratified lake, levels of NH_4^+ would only be elevated in the hypolimnion and levels of NO_3^- would only be elevated in the epilimnion. Both NH_4^+ and NO_3^- can be used as a nitrogen source by aquatic plants and algae. Total Kjeldahl nitrogen (TKN) is a measure of organic nitrogen plus ammonium. Adding the concentrations of TKN and nitrate together gives an indication of the amount of total nitrogen present in the water column. If inorganic nitrogen (NO_3^- , NO_2^- , NH_4^+) concentrations exceed 0.3 mg/L in spring, sufficient nitrogen is available to support summer algae blooms. However, low nitrogen levels do not guarantee limited algae growth the way low phosphorus levels do. Nitrogen gas in the air can dissolve in lake water and blue-green algae can "fix" atmospheric nitrogen, converting it into a usable form. Since other types of algae do not have the ability to do this, nuisance blue-green algae blooms are typically associated with lakes that are nitrogen limited (i.e., have low nitrogen levels).

The ratio of TKN plus nitrate nitrogen to total phosphorus (TN:TP) can indicate whether plant/algae growth in a lake is limited by nitrogen or phosphorus. Ratios of less than 10:1

suggest a system limited by nitrogen, while lakes with ratios greater than 20:1 are limited by phosphorus. It is important to know if a lake is limited by nitrogen or phosphorus because any addition of the limiting nutrient to the lake will, likely, result in algae blooms or an increase in plant density.

Solids:

Although several forms of solids (total solids, total suspended solids, total volatile solids, total dissolved solids) were measured each month by the Lakes Management Staff, total suspended solids (TSS) and total volatile solids (TVS) have the most impact on other variables and on the lake as a whole. TSS are particles of algae or sediment suspended in the water column. High TSS concentrations can result from algae blooms, sediment resuspension, and/or the inflow of turbid water, and are typically associated with low water clarity and high phosphorus concentrations in many lakes in Lake County. Low water clarity and high phosphorus concentrations, in turn, exacerbate the high TSS problem by leading to reduced plant density (which stabilize lake sediment) and increased occurrence of algae blooms. The median TSS value in epilimnetic waters in Lake County is 7.9 mg/L, ranging from below the 1 mg/L detection limit (10 lakes) to 165 mg/L in Fairfield Marsh.

TVS represents the fraction of total solids that are organic in nature, such as algae cells, tiny pieces of plant material, and/or tiny animals (zooplankton) in the water column. High TVS values indicate that a large portion of the suspended solids may be made up of algae cells. This is important in determining possible sources of phosphorus to a lake. If much of the suspended material in the water column is determined to be resuspended sediment that is releasing phosphorus, this problem would be addressed differently than if the suspended material was made up of algae cells that were releasing phosphorus. The median TVS value was 132 mg/L, ranging from 34 mg/L in Pulaski Pond to 298 mg/L in Fairfield Marsh.

Total dissolved solids (TDS) are the amount of dissolved substances, such as salts or minerals, remaining in water after evaporation. These dissolved solids are discussed in further detail in the *Alkalinity* and *Conductivity* sections of this document. TDS concentrations were measured in Lake County lakes prior to 2004, but was discontinued due to the strong correlation of TDS to conductivity and chloride concentrations.

Water Clarity:

Water clarity (transparency) is not a chemical property of lake water, but is often an indicator of a lake's overall water quality. It is affected by a lake's water color, which is a reflection of the amount of total suspended solids and dissolved organic chemicals. Thus, transparency is a measure of particle concentration and is measured with a Secchi disk. Generally, the lower the clarity or Secchi depth, the poorer the water quality. A decrease in Secchi depth during the summer occurs as the result of an increase in suspended solids (algae or sediment) in the water column. Aquatic plants play an important role in the level of water clarity and can, in turn, be negatively affected by low clarity levels. Plants increase clarity by competing with algae for

resources and by stabilizing sediments to prevent sediment resuspension. A lake with a healthy plant community will almost always have higher water clarity than a lake without plants. Additionally, if the plants in a lake are removed (through herbicide treatment or the stocking of grass carp), the lake will probably become dominated by algae and Secchi depth will decrease. This makes it very difficult for plants to become re-established due to the lack of available sunlight and the lake will, most likely, remain turbid. Turbidity will be accelerated if the lake is very shallow and/or common carp are present. Shallow lakes are more susceptible to sediment resuspension through wind/wave action and are more likely to experience clarity problems if plants are not present to stabilize bottom sediment.

Common Carp are prolific fish that feed on invertebrates in the sediment. Their feeding activities stir up bottom sediment and can dramatically decrease water clarity in shallow lakes. As mentioned above, lakes with low water clarity are, generally, considered to have poor water quality. This is because the causes and effects of low clarity negatively impact the plant and fish communities, as well as the levels of phosphorus in a lake. The detrimental impacts of low Secchi depth to plants has already been discussed. Fish populations will suffer as water clarity decreases due to a lack of food and decreased ability to successfully hunt for prey. Bluegills are planktivorous fish and feed on invertebrates that inhabit aquatic plants. If low clarity results in the disappearance of plants, this food source will disappear too. Largemouth Bass and Northern Pike are piscivorous fish that feed on other fish and hunt by sight. As the water clarity decreases, these fish species find it more difficult to see and ambush prey and may decline in size as a result. This could eventually lead to an imbalance in the fish community. Phosphorus release from resuspended sediment could increase as water clarity and plant density decrease. This would then result in increased algae blooms, further reducing Secchi depth and aggravating all problems just discussed. The average Secchi depth for Lake County lakes is 3.17 feet. From 2000-2005, Fairfield Marsh and Patski Pond had the lowest Secchi depths (0.33 feet) and Bangs Lake had the highest (29.23 feet). As an example of the difference in Secchi depth based on plant coverage, South Churchill Lake, which had no plant coverage and large numbers of Common Carp in 2003 had an average Secchi depth of 0.73 feet (over four times lower than the county average), while Deep Lake, which had a diverse plant community and few carp had an average 2003 Secchi depth of 12.48 feet (almost four times higher than the county average).

Another measure of clarity is the use of a light meter. The light meter measures the amount of light at the surface of the lake and the amount of light at each depth in the water column. The amount of attenuation and absorption (decreases) of light by the water column are major factors controlling temperature and potential photosynthesis. Light intensity at the lake surface varies seasonally and with cloud cover, and decreases with depth. The deeper into the water column light penetrates, the deeper potential plant growth. The maximum depth at which algae and plants can grow underwater is usually at the depth where the amount of light available is reduced to 0.5%-1% of the amount of light available at the lake surface. This is called the euphotic (sunlit) zone. A general rule of thumb in Lake County is that the 1% light level is about 1 to 3 times the Secchi disk depth.

Alkalinity, Conductivity, Chloride, pH:

Alkalinity:

Alkalinity is the measurement of the amount of acid necessary to neutralize carbonate (CO_3^{2-}) and bicarbonate (HCO_3^-) ions in the water, and represents the buffering capacity of a body of water. The alkalinity of lake water depends on the types of minerals in the surrounding soils and in the bedrock. It also depends on how often the lake water comes in contact with these minerals.

If a lake gets groundwater from aquifers containing limestone minerals such as calcium carbonate (CaCO_3) or dolomite (CaMgCO_3), alkalinity will be high. The median alkalinity in Lake County lakes (162 mg/L) is considered moderately hard according to the hardness classification scale of Brown, Skougstad and Fishman (1970). Because hard water (alkaline) lakes often have watersheds with fertile soils that add nutrients to the water, they usually produce more fish and aquatic plants than soft water lakes. Since the majority of Lake County lakes have a high alkalinity they are able to buffer the adverse effects of acid rain.

Conductivity and Chloride:

Conductivity is the inverse measure of the resistance of lake water to an electric flow. This means that the higher the conductivity, the more easily an electric current is able to flow through water. Since electric currents travel along ions in water, the more chemical ions or dissolved salts a body of water contains, the higher the conductivity will be. Accordingly, conductivity has been correlated to total dissolved solids and chloride ions. The amount of dissolved solids or conductivity of a lake is dependent on the lake and watershed geology, the size of the watershed flowing into the lake, the land uses within that watershed, and evaporation and bacterial activity. Many Lake County lakes have elevated conductivity levels in May, but not during any other month. This was because chloride, in the form of road salt, was washing into the lakes with spring rains, increasing conductivity. Most road salt is sodium chloride, calcium chloride, potassium chloride, magnesium chloride or ferrocyanide salts. Beginning in 2004, chloride concentrations are one of the parameters measured during the lake studies. Increased chloride concentrations may have a negative impact on aquatic organisms. Conductivity changes occur seasonally and with depth. For example, in stratified lakes the conductivity normally increases in the hypolimnion as bacterial decomposition converts organic materials to bicarbonate and carbonate ions depending on the pH of the water. These newly created ions increase the conductivity and total dissolved solids. Over the long term, conductivity is a good indicator of potential watershed or lake problems if an increasing trend is noted over a period of years. It is also important to know the conductivity of the water when fishery assessments are conducted, as electroshocking requires a high enough conductivity to properly stun the fish, but not too high as to cause injury or death.

pH:

pH is the measurement of hydrogen ion (H^+) activity in water. The pH of pure water is neutral at 7 and is considered acidic at levels below 7 and basic at levels above 7. Low pH levels of 4-5 are toxic to most aquatic life, while high pH levels (9-10) are not only toxic to aquatic life but may also result in the release of phosphorus from lake sediment. The presence of high plant densities can increase pH levels through photosynthesis, and lakes dominated by a large amount of plants or algae can experience large fluctuations in pH levels from day to night, depending on the rates of photosynthesis and respiration. Few, if any pH problems exist in Lake County lakes.

Typically, the flooded gravel mines in the county are more acidic than the glacial lakes as they have less biological activity, but do not usually drop below pH levels of 7. The median near surface pH value of Lake County lakes is 8.30, with a minimum of 7.06 in Deer Lake and a maximum of 10.28 in Round Lake Marsh North.

Eutrophication and Trophic State Index:

The word *eutrophication* comes from a Greek word meaning “well nourished.” This also describes the process in which a lake becomes enriched with nutrients. Over time, this is a lake’s natural aging process, as it slowly fills in with eroded materials from the surrounding watershed and with decaying plants. If no human impacts disturb the watershed or the lake, natural eutrophication can take thousands of years. However, human activities on a lake or in the watershed accelerate this process by resulting in rapid soil erosion and heavy phosphorus inputs. This accelerated aging process on a lake is referred to as cultural eutrophication. The term trophic state refers to the amount of nutrient enrichment within a lake system. *Oligotrophic* lakes are usually deep and clear with low nutrient levels, little plant growth and a limited fishery. *Mesotrophic* lakes are more biologically productive than oligotrophic lakes and have moderate nutrient levels and more plant growth. A lake labeled as *eutrophic* is high in nutrients and can support high plant densities and large fish populations. Water clarity is typically poorer than oligotrophic or mesotrophic lakes and dissolved oxygen problems may be present. A *hypereutrophic* lake has excessive nutrients, resulting in nuisance plant or algae growth. These lakes are often pea-soup green, with poor water clarity. Low dissolved oxygen may also be a problem, with fish kills occurring in shallow, hypereutrophic lakes more often than less enriched lakes. As a result, rough fish (tolerant of low dissolved oxygen levels) dominate the fish community of many hypereutrophic lakes. The categorization of a lake into a certain trophic state should not be viewed as a “good to bad” categorization, as most lake residents rate their lake based on desired usage. For example, a fisherman would consider a plant-dominated, clear lake to be desirable, while a water-skier might prefer a turbid lake devoid of plants. Most lakes in Lake County are eutrophic or hypereutrophic. This is primarily as a result of cultural eutrophication. However, due to the fertile soil in this area, many lakes (especially man-made) may have started out under eutrophic conditions and will never attain even mesotrophic conditions, regardless of any amount of money put into the management options. This is not an excuse to allow a lake to continue to deteriorate, but may serve as a reality check for lake owners attempting to create unrealistic conditions in their lakes.

The Trophic State Index (TSI) is an index which attaches a score to a lake based on its average

total phosphorus concentration, its average Secchi depth (water transparency) and/or its average chlorophyll *a* concentration (which represent algae biomass). It is based on the principle that as phosphorus levels increase, chlorophyll *a* concentrations increase and Secchi depth decreases. The higher the TSI score, the more nutrient-rich a lake is, and once a score is obtained, the lake can then be designated as oligotrophic, mesotrophic or eutrophic. Table 1 (below) illustrates the Trophic State Index using phosphorus concentration and Secchi depth.

Table 1. Trophic State Index (TSI).

Trophic State	TSI score	Total Phosphorus (mg/L)	Secchi Depth (feet)
Oligotrophic	<40	≤ 0.012	>13.12
Mesotrophic	$\geq 40 < 50$	$> 0.012 \leq 0.024$	$\geq 6.56 < 13.12$
Eutrophic	$\geq 50 < 70$	$> 0.024 \leq 0.096$	$\geq 1.64 < 6.56$
Hypereutrophic	≥ 70	> 0.096	< 1.64

**APPENDIX D. WATER QUALITY STATISTICS FOR ALL LAKE
COUNTY LAKES.**

2000 - 2007 Water Quality Parameters, Statistics Summary

	ALKoxic <=3ft00-2007		ALKanoxic 2000-2007	
Average	167.3		200	
Median	162.0		193	
Minimum	64.9	IMC	103	Heron Pond
Maximum	330.0	Flint Lake	470	Lake Marie
STD	42.0		48	
n =	803		253	

	Condoxic <=3ft00-2007		Condanoxic 2000-2007	
Average	0.8856		1.0035	
Median	0.8038		0.8340	
Minimum	0.2542	Broberg Marsh	0.3210	Lake Kathryn
Maximum	6.8920	IMC	7.4080	IMC
STD	0.5243		0.7787	
n =	802		252	

	NO3-N, Nitrate+Nitrite,oxic <=3ft00-2007		NH3-Nanoxic 2000-2007	
Average	0.515		2.070	
Median	0.150		1.340	
Minimum	<0.05	*ND	<0.1	*ND
Maximum	9.670	South Churchill Lake	18.400	Taylor Lake
STD	1.082		2.296	
n =	808		252	

*ND = Many lakes had non-detects (74.5%)

*ND = 19.8% Non-detects from 28 different lakes

Only compare lakes with detectable concentrations to the statistics above
Beginning in 2006, Nitrate+Nitrite was measured.

	pHoxic <=3ft00-2007		pHanoxic 2000-2007	
Average	8.31		7.22	
Median	8.31		7.21	
Minimum	7.07	Bittersweet #13	6.24	Banana Pond
Maximum	10.28	Round Lake Marsh	8.48	Heron Pond
STD	0.44	North	0.41	
n =	797		252	

	All Secchi 2000-2007	
Average	4.57	
Median	3.28	
Minimum	0.33	Fairfield Marsh, Patski Pond
Maximum	21.33	Bangs Lake
STD	3.81	
n =	750	



2000 - 2007 Water Quality Parameters, Statistics Summary (continued)

	TKNoxic <=3ft00-2007	
Average	1.457	
Median	1.220	
Minimum	<0.1	*ND
Maximum	10.300	Fairfield Marsh
STD	0.830	
n =	808	

*ND = 4.5% Non-detects from 16 different lakes

	TPoxic <=3ft00-2007	
Average	0.100	
Median	0.063	
Minimum	<0.01	*ND
Maximum	3.880	Albert Lake
STD	0.171	
n =	808	

*ND = 2.4% Non-detects from 7 different lakes
(Carina, Minear, & Stone Quarry)

	TSSall <=3ft00-2007	
Average	15.5	
Median	8.0	
Minimum	<0.1	*ND
Maximum	165.0	Fairfield Marsh
STD	20.3	
n =	814	

*ND = 1.8% Non-detects from 11 different lakes

	TDSoxic <=3ft00-2004	
Average	470	
Median	454	
Minimum	150	Lake Kathryn, White
Maximum	1340	IMC
STD	169	
n =	745	

No 2002 IEPA Chain Lakes.

	CLoxic <=3ft00-2007	
Average	211	
Median	158	
Minimum	30	White Lake
Maximum	2760	IMC
STD	247	
n =	411	

	TKNanoxic 2000-2007	
Average	2.910	
Median	2.320	
Minimum	<0.5	*ND
Maximum	21.000	Taylor Lake
STD	2.272	
n =	252	

*ND = 2.8% Non-detects from 4 different lakes

	TPanoxic 2000-2007	
Average	0.294	
Median	0.177	
Minimum	0.012	Indep. Grove and W. Loon Lake
Maximum	3.800	Taylor Lake
STD	0.380	
n =	252	

	TVSoxic <=3ft00-2007	
Average	135.3	
Median	132.0	
Minimum	34.0	Pulaski Pond Fairfield Marsh
Maximum	298.0	
STD	39.9	
n =	758	

No 2002 IEPA Chain Lakes

	CLanoxic <=3ft00-2007	
Average	232	
Median	119	
Minimum	41	Timber Lake (N)
Maximum	2390	IMC
STD	400	
n =	102	



77 of 163 lakes had anoxic conditions
Anoxic conditions are defined <=1 mg/l D.O.
pH Units are equal to the -Log of [H] ion activity
Conductivity units are in MilliSiemens/cm
Secchi Disk depth units are in feet
All others are in mg/L

Minimums and maximums are based on data from all lakes from 2000-2007 (n=1363).

Average, median and STD are based on data from the most recent water quality sampling year for each lake.

LCHD Lakes Management Unit ~ 12/17/2007

APPENDIX E. GRANT PROGRAM OPPORTUNITES.

Table E1. Potential Grant Opportunities

Grant Program Name	Funding Source	Contact Information	Funding Focus				Cost Share
			Water Quality/ Wetland	Habitat	Erosion	Flooding	
Challenge Grant Program	USFWS	847-381-2253 or 309-793-5800		X	X		
Chicago Wilderness Small Grants	CW	312-346-8166 ext. 30					None
Partners in Conservation (formerly C2000)	IDNR	http://dnr.state.il.us/orep/c2000/		X			None
Conservation Reserve Program	NRCS	http://www.nrcs.usda.gov/programs/crp/		X			Land
Ecosystems Program	IDNR	http://dnr.state.il.us/orep/c2000/ecosystem/		X			None
Emergency Watershed Protection	NRCS	http://www.nrcs.usda.gov/programs/ewp/			X	X	None
Five Star Challenge	NFWF	http://www.nfwf.org/AM/Template.cfm		X			None
Illinois Flood Mitigation Assistance Program	IEMA	http://www.state.il.us/iema/construction.htm				X	None
Great Lakes Basin Program	GLBP	http://www.glc.org/basin/stateproj.html?st=il	X		X		None
Illinois Clean Energy Community Foundation	ICECF	http://www.illinoiscleanenergy.org/		X			
Illinois Clean Lakes Program	IEPA	http://www.epa.state.il.us/water/financial-assistance/index.html					None
Lake Education Assistance Program (LEAP)	IEPA	http://www.epa.state.il.us/water/conservation-2000/leap/index.html	X				\$500

CW = Chicago Wilderness
 ICECF = Illinois Clean Energy Community Foundation
 IEMA = Illinois Emergency Management Agency
 IEPA = Illinois Environmental Protection Agency
 IDNR = Illinois Department of Natural Resources
 IDOA = Illinois Department of Agriculture
 LCSMC = Lake County Stormwater Management Commission
 LCSWCD = Lake County Soil and Water Conservation District
 NFWF = National Fish and Wildlife Foundation
 NRCS = Natural Resources Conservation Service
 USACE = United States Army Corps of Engineers
 USFWS = United States Fish and Wildlife Service

Table E1. Continued

Grant Program Name	Funding Source	Contact Information	Funding Focus				Cost Share
			Water Quality/ Wetland	Habitat	Erosion	Flooding	
Northeast Illinois Wetland Conservation Account	USFWF	847-381-2253	X				
Partners for Fish and Wildlife	USFWS	http://ecos.fws.gov/partners/		X			> 50%
River Network's Watershed Assistance Grants Program	River Network	http://www.rivernetwork.org	X	X	X		na
Section 206: Aquatic Ecosystems Restoration	USACE	312-353-6400, 309-794-5590 or 314-331-8404		X			35%
Section 319: Non-Point Source Management Program	IEPA	http://www.epa.state.il.us/water/financial-assistance/non-point.html	X	X			>40%
Section 1135: Project Modifications for the Improvement of the Environment	USACE	312-353-6400, 309-794-5590 or 314-331-8404		X			25%
Stream Cleanup And Lakeshore Enhancement (SCALE)	IEPA	http://www.epa.state.il.us/water/watershed/scale.html	X	X			None
Streambank Stabilization & Restoration (SSRP)	IDOA/ LCSWCD	http://www.agr.state.il.us/Environment/conserv/ or call LCSWCD at (847) 223-1056		X	X		25%
Watershed Management Boards	LCSMC	http://www.co.lake.il.us/smc/projects/wmb/default.asp	X		X	X	50%
Wetlands Reserve Program	NRCS	http://www.nrcs.usda.gov/programs/wrp/	X	X			Land
Wildlife Habitat Incentive Program	NRCS	http://www.nrcs.usda.gov/programs/whip/		X			Land

CW = Chicago Wilderness
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